

Solutions Manual to Accompany

Basic Principles and Calculations in Chemical Engineering

Seventh Edition

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PRENTICE HALL
Professional Technical Reference
Upper Saddle River, New Jersey 07458

TYPICAL EXAMS FOR A ONE SEMESTER COURSE (scheduled in the evening to avoid time constraints on students)

Exam No. 1 (Open Book, 1 1/2 hours)

PROBLEM 1 (35%)

Hydrogen can be separated from natural gas by diffusion through a metal tube. The rate of separation is given by

$$N = 2 \pi D_p R$$

where

- N = rate of transport of H_2 from the tube, g moles/(sec)(cm of length of tube)
- D = diffusion coefficient
- p = molar density of H_2 , g moles/cm³
- R = log mean radius of tube, $r_2 - r_1 / \ln \left(\frac{r_2}{r_1} \right)$, with r in cm.

What are the units of D ?

PROBLEM 2 (35%)

A pellet of bones weighing 10 tons is dropped from a lift truck from a height of 10 feet. The maximum velocity that pellet attains before hitting the ground is 8 ft/sec. How much kinetic energy does the pellet have in (ft)(lb) at this velocity?

PROBLEM 3 (25%)

The specific gravity of a fuel oil is 0.82. What is the density of the oil in lb/ft³? Show all units.

PROBLEM 4 (35%)

Sulfur trioxide (SO_3) can be absorbed in sulfuric acid solution to form more concentrated sulfuric acid. If the gas to be absorbed contains 10% SO_3 , 40% H_2 , 3% SO_2 , and 47% O_2 , how many parts per million of O_2 are there in the gas? (b) What is the composition of the gas on a H_2 free basis?

PROBLEM 5 (35%)

You have 100 kilograms of gas of the following composition:

CH_4	30%
H_2	10%
N_2	60%

What is the average molecular weight of this gas?

PROBLEM 6 (25%)

If the heat capacity of a substance is 3.32 Btu/(°C) and its molecular weight is 25.4, what is its heat capacity in

- (a) Btu/(°F)
- (b) Btu/(°C)
- (c) Btu/(lb)(°C)

PROBLEM 7 (35%)

A rock containing 100% $BaSO_4$ is heated with coke (94% C, 6% ash) and the composition of the product is $BaSO_4$ (11.0%), BaS (70.9%), C (11.9%), ash (2.2%). The reaction is



Calculate the percent excess reactant, and the degree of completion of the reaction.

PROBLEM 8 (35%)

A gas cylinder to which is attached an Bourdon gage appears to be at a pressure of 21.38 in. Hg at 70°F. The thermometer reads 101.8 kPa. A student claims that the pressure in the tank is 1.3 psia, but another student points out that this is impossible - the pressure is really 28.2 psia. Can 1.3 psia be correct? Explain and show calculations to back up your explanation.

Solution Manual For Himmelblau

Ying Liu



Solution Manual For Himmelblau:

Solution Manual For Himmelblau Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Solution Manual For Himmelblau**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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