



Interpreting Solubility Curve



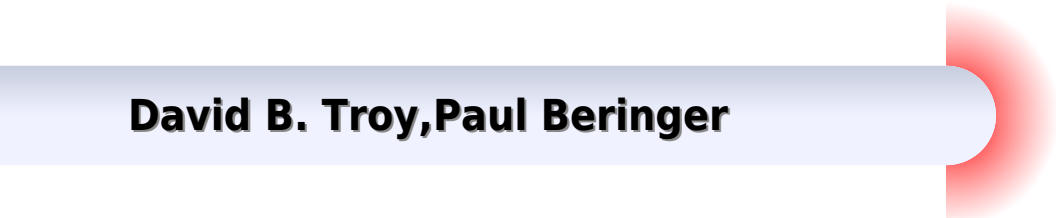
1. Define solubility. _____
2. Using the solubility curve below, answer the given questions.



- I. What is the solvent in this solubility curve? _____
- II. Saturation points are given per how much of the solvent? _____
- III. What is the saturation point of NaNO₃ at 10°C, 40°C, & 80°C?
10°C = _____ 40°C = _____ 80°C = _____
- IV. How much NH₃ can you dissolve in water at 10°C, 30°C, & 90°C?
10°C = _____ 30°C = _____ 90°C = _____
- V. Which salt is the least soluble in water? _____
- VI. How many grams of potassium chloride can be dissolved in 200 g of water at 80 °C? _____
- VII. At 40 °C, how much potassium nitrate will be dissolved in 300 g of water? _____
- VIII. Which salt shows the least change in solubility from 0° to 100° C? _____
- IX. Which substance shows a decrease in solubility from 0° to 100° C? _____
- X. At 30 °C, 90 g of sodium nitrate is dissolved in 100 g water. Is this solution saturated, unsaturated, or supersaturated? _____

Solubility Curve Practice Problems Packet Answers

David B. Troy, Paul Beringer



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In some sort of consumed by displays and the ceaseless chatter of fast connection, the melodic elegance and mental symphony created by the written term often disappear in to the back ground, eclipsed by the relentless sound and disruptions that permeate our lives. Nevertheless, situated within the pages of **Solubility Curve Practice Problems Packet Answers** a charming fictional treasure overflowing with fresh feelings, lies an immersive symphony waiting to be embraced. Constructed by a masterful musician of language, that fascinating masterpiece conducts viewers on a mental trip, well unraveling the concealed tunes and profound influence resonating within each carefully crafted phrase. Within the depths of this poignant analysis, we can explore the book is key harmonies, analyze its enthralling writing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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