

Periodic Table WORKSHEET

Practice Using the Periodic Table

Name _____

Directions: Use the Periodic Table to find the following information.

1. Find the atomic number of:

hydrogen _____	potassium _____	lead _____	aluminum _____
oxygen _____	barium _____	iodine _____	krypton _____
cobalt _____	iron _____	calcium _____	tin _____
vanadium _____	cadmium _____	polonium _____	iodine _____
sodium _____	strontium _____	rhodium _____	vanadium _____
neon _____	antimony _____	cesium _____	oganesson _____

2. Find the atomic mass of (round off to nearest whole number):

beryllium _____	oxygen _____	iron _____	calcium _____
aluminum _____	iodine _____	barium _____	phosphorus _____
cadmium _____	hydrogen _____	nitrogen _____	silver _____
gallium _____	tin _____	rubidium _____	europium _____
copper _____	thallium _____	manganese _____	vanadium _____

3. Find the following element names:

Li _____	Ce _____
Cl _____	Sn _____
Si _____	V _____
Fe _____	Ba _____
	Be _____

4. Find the following element symbols:

potassium _____	nitrogen _____
hydrogen _____	calcium _____
silver _____	iron _____
arsenic _____	krypton _____
polonium _____	barium _____
cesium _____	berkelium _____

5. Find the following element names:

beryllium _____	berkium _____
chromium _____	mercury _____
francium _____	phosphorus _____
hydrogen _____	nickel _____
boron _____	iron _____
manganese _____	oganesson _____

6. Find the following element symbols:

fluorine _____	iron _____
nitrogen _____	silver _____
rubidium _____	europium _____
aluminum _____	lead _____

Using The Periodic Table Answer Key

Chandan Sengupta



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offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life **Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28

The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement *Chemistry in the Community (ChemCom)*

American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today's teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards **Literacy in Science and Technology, Grades 6 - 8** Schyrlet Cameron, Suzanne Myers, 2014-01-15

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standards Academic Chemistry IX , **CBSE - ICSE Chemistry Part I** Chandan Sengupta, First Publication October 2021 Place of Publication Arabinda Nagar Bankura 722101 This workbook will provide an ample scope in getting exposed to the system of acquiring skills and competence related to the understanding of chemistry It also exposes the student to the concepts of chemistry for enabling the aspirant in acquisition of skills related to chemistry Some of the worksheets are prepared along with model answers Some other worksheets are meant for self assessment and evaluation purposes It is also observed that some of the topics are specific to the referred curriculum Some other topics are varyingly incorporated in other streams of study Culmination of more than two streams will enable the fellow student to cope up with the preparatory works meant for Olympiads and other competitive examinations

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