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Chapter 5 The Periodic Rable

Investigation SA

Using Clues to Identify Elements

Background Information

Chemical elements can be classified according to their properties as metals, normetals, and metalloids. Metals are good conductors of heat and electricity. Many metals are malliable and ductile. Nonmetals are poor conductors of heat and electricity, and solid normetals tend to be brittle. Metalloids have properties between those of metals and nonmetals.

Elements in the same group on the periodic table have the same number of valence electrons, which are electrons in the highest occupied energy level of an atom. The number of an A group matches the number of valence electrons in atoms of each element in the group. For example, the Group 4A elements each have four valence electrons per atom. The exception to this pattern is the element helium, which is in Group 8A leat has only two valence electrons.

Because elements in a group have the same reamber of valence electrons, they tend to have similar properties. The most reactive metals are the alkali metals in Group IA on the far left side of the periodic table. The Group IA elements are the alkaline earth metals, which are somewhat less reactive than the alkali metals. Groups of elements become less metallic in their properties from left to right across the periodic table. The most reactive more from left to right across the periodic table. The most reactive more from the truth are the colories and odorless, and narely react with other elements.

In this investigation, you will use a list of closes to identify 34 elements and place them in their correct locations in the periodic table.

Problem

Where do the elements described in the class fit in the periodic table?

Pre-Lab Discussion

Bend the entire investigation. Then, work with a partner to answer the following questions.

Analyzing Data. How will the index cards be useful when you are reading the clues to the elements?		
-		

Using Clues To Identify Elements Lab Key

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Using Clues To Identify Elements Lab Key:

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the professor and students agreed to construct a class that represented a reflective and grounded experience in the policy analysis of a current and relevant issue with global ramifications we began exploring policies that were developed at the global level and that were implemented locally We investigated the surge of globally developed standards and regulations in an effort to improve education Our goal was to learn cross nationally about policies that seek to reform curriculum and instruction under efficiency and global competitiveness arguments such as Education for All EFA and its USA cousin No Child Left Behind NCLB We knew our work would be bounded by the time available in a one semester class and by resource constraints We did exploratory inquiry supported by literature reviews reports on rigorous research studies and in one case an exploratory case study The policies we chose to explore such as EFA and NCLB offered us the opportunity to examine current reform tendencies that are intended to provide access to quality education for all children the preparation of teachers to support diverse populations the organization of schools to accommodate these children in response to vague policy mandates and power issues affecting the different constituencies and stakeholders. The effects of these and other policies were difficult to track because research is scant and decisions are frequently made based on ideology or political persuasion Our purpose was to explore the critical issues that originated such policies and to search for documented evidence regarding policy implementation and effectiveness We investigated the factors that seemed to interfere with successful implementation from conceptual theoretical and methodological perspectives In this class we learned that there are not ready set frameworks for policy analysis but rather that these have to be constructed according to the issues that emerge as policies are conceptualized and implemented to fit local contexts and needs The book pays particular attention to the contexts of policy including the evolving conceptualization of global and local systems of governance knowledge regimes and policy spaces The book is designed for faculty and doctoral students in education who are interested in understanding diverse frameworks for policy analysis and for those in the general public who are interested in the policies we analyze here

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a pneumatic test poses a significant hazard due to the amount of stored energy in the compressed gas a rupture could result in an explosive release of energy for example an explosion of a 200 ft 60 m of 36 in pipe containing 500 psi 3500 kpa can create a blast wave nearly equivalent to 80 lb 35 kg of tnt

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