

The Advanced Placement Examination in Chemistry

Part I – Multiple Choice Questions

Part II - Free Response Questions

Selected Questions from 1970 to 2010

Stoichiometry

Part I

1984

2. Which of the following forms a compound having the formula KNO_3 ?
- (A) F
(B) S
(C) Mg
(D) Ar
(E) Mn
32. The net ionic equation for the reaction between silver carbonate and hydrochloric acid is
- (A) $\text{Ag}_2\text{CO}_3(s) + 2 \text{H}^+ + 2 \text{Cl}^- \rightarrow 2 \text{AgCl}(s) + \text{H}_2\text{O} + \text{CO}_2(g)$
(B) $2 \text{Ag}^+ + \text{CO}_3^{2-} + 2 \text{H}^+ + 2 \text{Cl}^- \rightarrow 2 \text{AgCl}(s) + \text{H}_2\text{O} + \text{CO}_2(g)$
(C) $\text{CO}_3^{2-} + 2 \text{H}^+ \rightarrow \text{H}_2\text{O} + \text{CO}_2(g)$
(D) $\text{Ag}^+ + \text{Cl}^- \rightarrow \text{AgCl}(s)$
(E) $\text{Ag}_2\text{CO}_3(s) + 2 \text{H}^+ \rightarrow 2 \text{Ag}^+ + \text{H}_2\text{CO}_3$
44. What number of moles of O_2 is needed to produce 14.2 grams of P_2O_5 from P? (Molecular weight $\text{P}_2\text{O}_5 = 284$)
- (A) 0.0500 mole
(B) 0.0625 mole
(C) 0.125 mole
(D) 0.250 mole
(E) 0.500 mole
45. The alkenes are compounds of carbon and hydrogen with the general formula C_nH_{2n} . If 0.561 gram of any alkene is burned in excess oxygen, what number of moles of H_2O is formed?
- (A) 0.0400 mole
(B) 0.0600 mole
(C) 0.0800 mole
(D) 0.400 mole
(E) 0.800 mole
52. $3 \text{Ag}(s) + 4 \text{HNO}_3 \rightarrow 3 \text{AgNO}_3 + \text{NO}(g) + 2 \text{H}_2\text{O}$
The reaction of silver metal and dilute nitric acid proceeds according to the equation above. If 0.10 mole of powdered silver is added to 10. milliliters of 6.0-molar nitric acid, the number of moles of NO gas that can be formed is
- (A) 0.015 mole
(B) 0.020 mole
(C) 0.030 mole
(D) 0.045 mole
(E) 0.090 mole

Solution Stoichiometry Multiple Choice Questions

**Peter J. Mikulecky, Michelle Rose
Gilman, Kate Brutlag**



Solution Stoichiometry Multiple Choice Questions:

Multiple Solution Methods for Teaching Science in the Classroom Stephen DeMeo, 2008 For the first time in science education the subject of multiple solution methods is explored in book form While a multiple method teaching approach is utilized extensively in math education there are very few journal articles and no texts written on this topic in science Teaching multiple methods to science students in order to solve quantitative word problems is important for two reasons First it challenges the practice by teachers that one specific method should be used when solving problems Secondly it calls into question the belief that multiple methods would confuse students and retard their learning Using a case study approach and informed by research conducted by the author this book claims that providing students with a choice of methods as well as requiring additional methods as a way to validate results can be beneficial to student learning A close reading of the literature reveals that time spent on elucidating concepts rather than on algorithmic methodologies is a critical issue when trying to have students solve problems with understanding It is argued that conceptual understanding can be enhanced through the use of multiple methods in an environment where students can compare evaluate and verbally discuss competing methodologies through the facilitation of the instructor This book focuses on two very useful methods proportional reasoning PR and dimensional analysis DA These two methods are important because they can be used to solve a large number of problems in all of the four academic sciences biology chemistry physics and earth science This book concludes with a plan to integrate DA and PR into the academic science curriculum starting in late elementary school through to the introductory college level A challenge is presented to teachers as well as to textbook writers who rely on the single method paradigm to consider an alternative way to teach scientific problem solving *Oswaal NCERT Exemplar (Problems - Solutions) Class 11 Physics, Chemistry and Biology (Set of 3 Books) For 2024 Exam* Oswaal Editorial Board, 2023-10-28 Description of the product Chapter wise and Topic wise presentation Chapter wise Objectives A sneak peek into the chapter Mind Map A single page snapshot of the entire chapter Revision Notes Concept based study materials Tips Tricks Useful guidelines for attempting each question perfectly Some Commonly Made Errors Most common and unidentified errors are focused Expert Advice Oswaal Expert Advice on how to score more Oswaal QR Codes For Quick Revision on your Mobile Phones and Tablets *Oswaal NCERT Exemplar (Problems - Solutions) Class 11 Physics, Chemistry and Mathematics (Set of 3 Books) For 2024 Exam* Oswaal Editorial Board, 2023-10-28 Description of the product Chapter wise and Topic wise presentation Chapter wise Objectives A sneak peek into the chapter Mind Map A single page snapshot of the entire chapter Revision Notes Concept based study materials Tips Tricks Useful guidelines for attempting each question perfectly Some Commonly Made Errors Most common and unidentified errors are focused Expert Advice Oswaal Expert Advice on how to score more Oswaal QR Codes For Quick Revision on your Mobile Phones and Tablets **A Level Chemistry MCQ (Multiple Choice Questions)** Arshad Iqbal, 2019-06-18 The A Level Chemistry Multiple Choice Questions

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