QUIZIZZ X

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Unit 8
Acids & Bases

2.

8.1 Introduction to Acids & Bases

Autologization of Water:



This is an ensisting the process. Increasing the best will shift the resolion to produce more E.O. and OF. X. increases imore products, Sorth pill sent pill dearman but pure II, O will will be resolion.

Kw = [H,O'][OH]

Only digits to the right of the decimal are significant in pill and pOH value. He

In case of oridic solution > [H⁺] > [OH⁺]

[In case of neutral solution > [H⁺] = [OH⁻]

In case of back solution > [H*] < [OH*]

ON ROBERTSON SHIERT:

$$K_{ac} = \text{HP BOH } 1 = 1.0 \times 10^{-15} \text{ at 25°C}$$

= $K_{ac} \times K_{bc}$

$$pH = -log(H^*)$$
, $pOH = -log(OH)$

$$HOq - Hq = 0.0$$

Drill-on and 10001-on-

The value of K_m for water at 0°C is 1×10° ¹⁰. What is the pOH of water at 0°C?

40 7.5

60 80

49 7:0

40, 655

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CHANG

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