

INTRODUCTION

INTRODUCTION

GENERAL INFORMATION

This shop manual covers the following BRP made 2008 REV XP series models:

MODEL	ENGINE TYPE
GSX Limited 600 HO SDI	593
GSX Limited 800R Power TEK	797
GSX Limited Touring 600 HO SDI	593
GSX Sport 500SS	593
MX Z Adrenaline 600 HO SDI	593
MX Z Adrenaline 800R Power TEK	797
MX Z Renegade 600 HO SDI	593
MX Z Renegade 800R Power TEK	797
MX Z Renegade X 600 HO SDI	593
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MX Z TNT 500SS	593
MX Z TRAIL 500SS	593
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MX Z X 800R Power TEK	797
SUMMIT EVEREST 800R Power TEK	797
SUMMIT X 800R Power TEK	797

The information and component/system descriptions contained in this manual are correct at time of publication. BRP however, maintains a policy of continuous improvement of its products without imposing upon itself any obligation to install them on products previously manufactured.

Due to late changes, it may have some differences between the manufactured product and the description and/or specifications in this document.

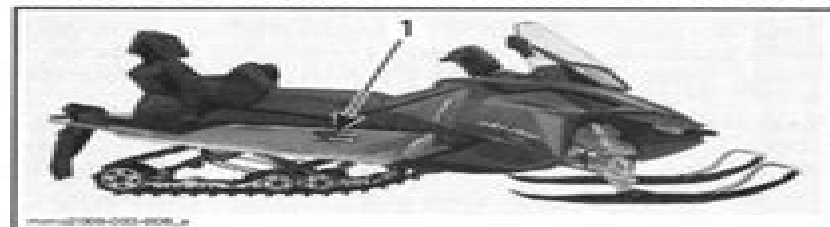
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This manual uses technical terms which may be slightly different from the ones used in the *PARTS CATALOG*.

When ordering parts always refer to the specific model *PARTS CATALOG*.

VEHICLE IDENTIFICATION NUMBER

Vehicle Identification Number Location



TYPICAL
1. Vehicle identification number

Identification Number Meaning

2	B	P	S	B	X	5	A	9	Y	1	0	0	0	0	1
Model number											Serial number				
Model year: y = 2000															
											1 = 2001				
											2 = 2002				
											etc.				

ENGINE SERIAL NUMBER

Engine Serial Number Location



TYPICAL — 500SS AND 600 HO SDI ENGINES
1. Engine serial number

Ski Doo Mxz 800r Power Tek 2008 Service Manual

Haynes Publishing



Ski Doo Mxz 800r Power Tek 2008 Service Manual:

Reviewing **Ski Doo Mxz 800r Power Tek 2008 Service Manual**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Ski Doo Mxz 800r Power Tek 2008 Service Manual**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k

for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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