

3.3 Thermodynamics: Study Guide

Use this [outline](#) to review the key vocabulary for this unit.

1. Match the vocabulary words on the left with the appropriate definitions and units on the right.

a. Thermal Energy	<u>A</u> Kinetic energy in transit from one object to another due to temperature difference
b. Temperature	<u>B</u> Degree Celsius
	<u>C</u> Joules
	<u>D</u> The average kinetic energy of particles in an object

2. In the table below, write in the appropriate formulas.

Energy Transfer	Rate of Energy Transfer	Thermal Conductivity	Coefficient of Heat Conductivity
$Q = mc \cdot \Delta T$	$P = \frac{c \Delta T}{L}$ or $P = \frac{Q}{t}$ <u>Thermal conductance</u> $P = \sigma A c (\Delta T)^2$	$K = \frac{PL}{A \Delta T}$	$U = \frac{P}{A \cdot \Delta T}$

3. For each example, write the correct number for the law of thermodynamics it describes (0, 1, 2).
- 2 The law of entropy
 - 1 Thermal energy can only be increased in a system by adding heat from an external source or performing work within the system
 - 0 If two systems are separately found to be in thermal equilibrium with a third, then the two systems are in thermal equilibrium with each other
 - 2 Thermal energy flows only from warmer to cooler
 - 1 The law of conservation of energy applied to thermal systems
4. How much thermal energy would it take to raise the temperature of 1.1 (1 kg) of water by 10°C? The specific heat capacity of water is $4184 \frac{J}{kg \cdot ^\circ C}$

$$Q = mc \Delta T$$

$$= 1.1 kg \cdot 4184 \frac{J}{kg \cdot ^\circ C} \cdot 10^\circ C$$

$$Q = 41,840 J$$

$$Q = \underline{41,840 J}$$

5. A materials scientist is trying to determine the specific heat capacity of an experimental material. She applies 30 kJ of thermal energy to a 5 kg sample of the material and records a 24°C temperature increase in the sample. What is the specific heat capacity of the material?

$$Q = mc \Delta T$$

$$c = \frac{Q}{m \Delta T}$$

$$c = 2500 \frac{J}{kg \cdot ^\circ C}$$

$$c = \underline{2500 \frac{J}{kg \cdot ^\circ C}}$$

$$= \frac{30000 J}{5 kg \cdot 24^\circ C}$$

Thermodynamics Study Guide Answers

Arshad Iqbal



Thermodynamics Study Guide Answers:

Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

Study Guide/Solutions Manual for Organic Chemistry K. Peter C. Vollhardt, Neil E. Schore, 2018-01-12 Updated for the Eighth Edition of Vollhardt Schore Organic Chemistry and written by the book's coauthor Neil Schore this invaluable manual includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

Student Study Guide/solutions Manual to Accompany Foundations of Inorganic, Organic, and Biological Chemistry Katherine J. Denniston, 1995 This student study guide has been written to accompany Foundations of Inorganic Organic and Biological Chemistry by Caret Denniston and Topping It was designed to complement the text not to be used in place of the text Each chapter of the study guide contains the following sections reorganized and expanded set of learning goals concise chapter summary in chapter solved problems list of key terms review problems Preface

Thermodynamics Ibrahim Dincer, 2020-08-18 Presents a unique stepwise exergy based approach to thermodynamic concepts systems and applications Thermodynamics A Smart Approach redefines this crucial branch of engineering as the science of energy and exergy rather than the science of energy and entropy to provide an innovative step by step approach for teaching understanding and practicing thermodynamics in a clearer and easier way Focusing primarily on the concepts and balance equations this innovative textbook covers exergy under the second law of thermodynamics discusses exergy matters and relates thermodynamics to environmental impact and sustainable development in a clear simple and understandable manner It aims to change the way thermodynamics is taught and practiced and help overcome the fear of thermodynamics Author Ibrahim Dincer a pioneer in the areas of thermodynamics and sustainable energy technologies draws upon his multiple decades of experience teaching and researching thermodynamics to offer a unique exergy based approach to the subject Enabling readers to easily comprehend and apply thermodynamic principles the text organizes thermodynamics into seven critical steps property state process cycle first law of thermodynamics second law of thermodynamics and performance assessment and provides extended teaching tools for systems and applications Precise student friendly chapters cover fundamental concepts thermodynamic laws conventional and innovative power and refrigeration cycles and more This textbook Covers a unique approach in teaching design analysis and assessment of thermodynamic systems Provides lots of examples for every subject for students and instructors Contains hundreds of

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Biomass as a Sustainable Energy Source for the Future Wiebren de Jong, J. Ruud van Ommen, 2014-10-03 Focusing on the conversion of biomass into gas or liquid fuels the book covers physical pre treatment technologies thermal chemical and biochemical conversion technologies Details the latest biomass characterization techniques Explains the biochemical and thermochemical conversion processes Discusses the development of integrated biorefineries which are similar to petroleum refineries in concept covering such topics as reactor configurations and downstream processing Describes how to mitigate the environmental risks when using biomass as fuel Includes many problems small projects sample calculations and industrial application examples

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Select Thermodynamic Models for Process Simulation
Jean-Charles de Hemptinne, Jean-Marie Ledanois, 2012 The selection of the most adequate thermodynamic model in a process simulation is an issue that most process engineer has to face sooner or later This book conceived as a practical guide aims at providing adequate answers by analysing the questions to be looked at The analysis first chapter yields three keys that are further discussed in three different chapters 1 A good understanding of the properties required in the process and their method of calculation is the first key The second chapter provides to that end in a synthetic manner the most important equations that are derived from the fundamental principles of thermodynamics 2 An adequate description of the mixture which is a combination of models and parameters is the second key The third chapter makes the link between components and models both from a numerical parameterisation and physical molecular interactions point of view Finally 3 a correct view of the phase behaviour and trends in regard of the process conditions is the third key The fourth chapter illustrates the phase behaviour and makes model recommendations for the most significant industrial systems A decision tree is provided at the end of this chapter In the last chapter the key questions are reviewed for a number of typical processes This book is intended for process engineers who are not specialists of thermodynamics but are confronted with this kind of problems and need a reference book as well as process engineering students who will find an original approach to thermodynamics complementary of traditional lectures

Introduction to Thermodynamics and the Zeroth Law. Internal Energy and the First Law on Thermodynamics Open University, 1973

Study Guide for the Professional Licensure of Mining and Mineral Processing Engineers Society for Mining, Metallurgy, and Exploration (U.S.), 2008 This handy workbook prepares you for professional licensure and allows you to practice your test taking skills The text covers the history of professional licensure

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