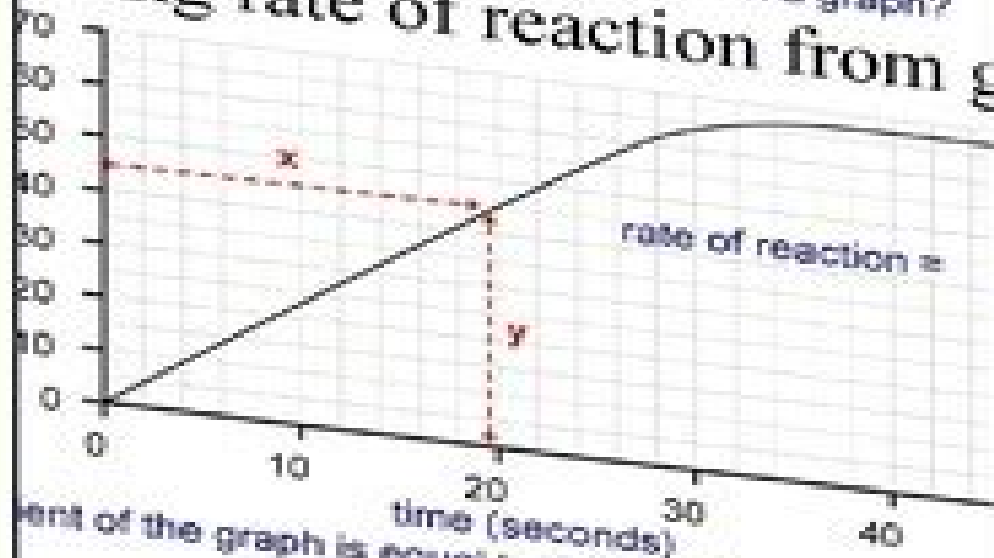


THERMOCHEMISTRY

Calculating rate of reaction from



time (seconds)

40

30

20

10

0

0 10 20 30 40 50 60 70 80 90 100

percentage of the graph is equal to the initial rate of reaction

section = 45

reaction = $\frac{45\text{cm}^3}{20\text{s}}$

ION REACTIONS

ut, as products form, it speeds up
and there will come a time when t
al and opposite; the reaction has



PLANTING
AND



**FORWARD REACT
AS REACTANT**



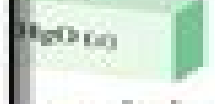
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low into or
pressure.
(stants)

at constant



Consider the
secret value
by which we
know that



REACTIONS

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LM Reder

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