

THERMODYNAMICS TEST REVIEW ANSWERS

Specific Heat Problems

1. How many J would it take to raise the temperature of 200 grams of water from 5 degree Celsius C to 85 degree Celsius?

Answer: $Q = (200 \text{ g})(4.18)(80 \text{ }^{\circ}\text{C}) = 66880 \text{ J}$

2. How many J would problem number 1 be if it was aluminum instead of water?

Answer: $Q = (200 \text{ g})(0.897)(80 \text{ }^{\circ}\text{C}) = 14352 \text{ J}$

3. How many grams of copper could be heated from 20 °C to 75 °C if 1200J are applied to it?

Answer: $1200 \text{ J} = (x)(0.385)(55 \text{ }^{\circ}\text{C}) = 56.7 \text{ g}$

4. What is the specific heat capacity of a substance if 750 J caused 100 grams of it to go from 90 °C to 135 °C?

Answer: $750 \text{ J} = (100 \text{ g})(x)(45 \text{ }^{\circ}\text{C}) = 0.167 \text{ J/g}^{\circ}\text{C}$

5. What would the final temperature be if 500 J are applied to 150 grams of ice at -90 °C?

Answer: $500 \text{ J} = (150 \text{ g})(2.06)(x - (-90 \text{ }^{\circ}\text{C})) = -88.4 \text{ }^{\circ}\text{C}$

6. What would the temperature change by if a 90 gram piece of hot iron cooled by losing 200 J.

Answer: $-200 \text{ J} = (90 \text{ g})(0.449)(x) = -4.95 \text{ }^{\circ}\text{C}$

7. What was the initial temperature if 500 J were applied to 250 g of mercury and the final temperature was 50 °C?

Answer: $500 \text{ J} = (250 \text{ g})(0.140)(50 \text{ }^{\circ}\text{C} - x) = 35.7 \text{ }^{\circ}\text{C}$

Calorimetry

1. What is the specific heat capacity of water?

Answer: $4.18 \text{ J/g}^{\circ}\text{C}$

2. A piece of ice is placed on the sidewalk on a hot summer day. Describe what happens to energy content of the piece of ice, does it gain or lose energy? What happens to the air around the ice, does it gain or lose energy?

Answer: The ice gains energy from the surrounding air and increases its energy (endothermic). The air loses energy to the ice and decreases its energy (exothermic)

Thermodynamics Test Review Answers

IM Harris



Thermodynamics Test Review Answers:

This book delves into Thermodynamics Test Review Answers. Thermodynamics Test Review Answers is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Thermodynamics Test Review Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Thermodynamics Test Review Answers
 - Chapter 2: Essential Elements of Thermodynamics Test Review Answers
 - Chapter 3: Thermodynamics Test Review Answers in Everyday Life
 - Chapter 4: Thermodynamics Test Review Answers in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, the author will provide an overview of Thermodynamics Test Review Answers. This chapter will explore what Thermodynamics Test Review Answers is, why Thermodynamics Test Review Answers is vital, and how to effectively learn about Thermodynamics Test Review Answers.
3. In chapter 2, the author will delve into the foundational concepts of Thermodynamics Test Review Answers. This chapter will elucidate the essential principles that need to be understood to grasp Thermodynamics Test Review Answers in its entirety.
4. In chapter 3, the author will examine the practical applications of Thermodynamics Test Review Answers in daily life. The third chapter will showcase real-world examples of how Thermodynamics Test Review Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Thermodynamics Test Review Answers in specific contexts. The fourth chapter will explore how Thermodynamics Test Review Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Thermodynamics Test Review Answers. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Thermodynamics Test Review Answers.

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