

- C. repulsion of molecules
- D. attraction of molecules
- E. surface tension of molecules.

Answer: B

16. According to kinetic theory of gases, the absolute zero temperature is attained when

- A. volume of the gas is zero
- B. pressure of the gas is zero
- C. kinetic energy of the molecules is zero
- D. specific heat of gas is zero
- E. mass is zero.

Answer: C

17. Kinetic theory of gases assumes that the collisions between the molecules are

- A. perfectly elastic
- B. perfectly inelastic
- C. partly elastic
- D. partly inelastic
- E. partly elastic and partly inelastic.

Answer: A

18. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to

- A. E/3
- B. E/2
- C. 2E/3
- D. 3E/2
- E. 3E/4

Answer: A

19. Kinetic energy of the molecules in terms of absolute temperature T is proportional

- A. T
- B. T^2
- C. $T^{1/2}$
- D. $T^{3/2}$
- E. $T^{5/2}$

Answer: A

20. Supercooled vapour behaves

- A. exactly as gas
- B. as liquid
- C. as solid
- D. as solid-liquid vapour
- E. supercooled as liquid

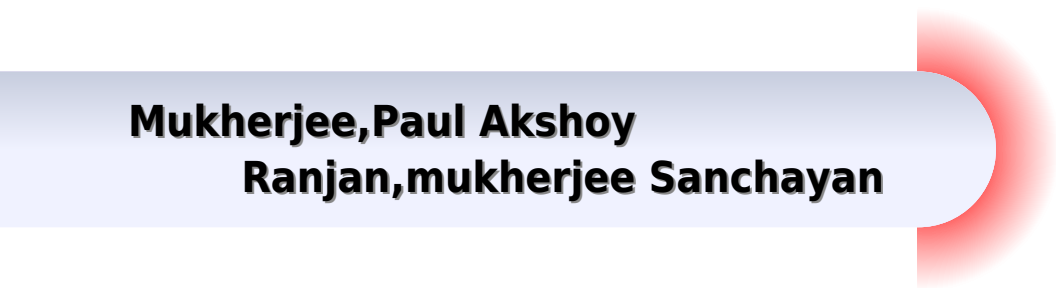
Answer: A

21. Absolute zero pressure will occur

- A. at absolute zero
- B. at the centre of the earth

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Mukherjee,Paul Akshoy
Ranjan,mukherjee Sanchayan



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Rotational Motion MCQ Chapter 29 Second Law of Thermodynamics MCQ Chapter 30 Simple Harmonic Motion MCQ Chapter 31 Special Relativity MCQ Chapter 32 Straight Line Motion MCQ Chapter 33 Transverse Waves MCQ Chapter 34 Two and Three Dimensional Motion MCQ Chapter 35 Vector Quantities MCQ Chapter 36 Work Kinetic Energy Theorem MCQ The Alternating Fields and Currents MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Alternating current damped oscillations in an RLS circuit electrical mechanical analog forced and free oscillations LC oscillations phase relations for alternating currents and voltages power in alternating current circuits transformers The Astronomical Data MCQ PDF e Book Chapter 2 practice test to solve MCQ questions on Aphelion distance from earth eccentricity of orbit equatorial diameter of planets escape velocity of planets gravitational acceleration of planets inclination of orbit to earth s orbit inclination of planet axis to orbit mean distance from sun to planets moons of planets orbital speed of planets perihelion period of rotation of planets planet densities planets masses sun earth and moon The Capacitors and Capacitance MCQ PDF e Book Chapter 3 practice test to solve MCQ questions on Capacitor in parallel and in series capacitor with dielectric charging a capacitor cylindrical capacitor parallel plate capacitor The Circuit Theory MCQ PDF e Book Chapter 4 practice test to solve MCQ questions on Loop and junction rule power series and parallel resistances single loop circuits work energy and EMF The Conservation of Energy MCQ PDF e Book Chapter 5 practice test to solve MCQ questions on Center of mass and momentum collision and impulse collisions in one dimension conservation of linear momentum conservation of mechanical energy linear momentum and Newton s second law momentum and kinetic energy in collisions Newton s second law for a system of particles path independence of conservative forces work and potential energy The Coulomb s Law MCQ PDF e Book Chapter 6 practice test to solve MCQ questions on Charge is conserved charge is quantized conductors and insulators and electric charge The Current Produced Magnetic Field MCQ PDF e Book Chapter 7 practice test to solve MCQ questions on Ampere s law and law of Biot Savart The Electric Potential Energy MCQ PDF e Book Chapter 8 practice test to solve MCQ questions on Introduction to electric potential energy electric potential and equipotential surfaces The Equilibrium Indeterminate Structures MCQ PDF e Book Chapter 9 practice test to solve MCQ questions on Center of gravity density of selected materials of engineering interest elasticity equilibrium indeterminate structures ultimate and yield strength of selected materials of engineering interest and Young s modulus of selected materials of engineering interest The Finding Electric Field MCQ PDF e Book Chapter 10 practice test to solve MCQ questions on Electric field electric field due to continuous charge distribution electric field lines flux and Gauss law The First Law of Thermodynamics MCQ PDF e Book Chapter 11 practice test to solve MCQ questions on Absorption of heat by solids and liquids Celsius and Fahrenheit scales coefficients of thermal expansion first law of thermodynamics heat of fusion of common substances heat of transformation heat of vaporization of common substances introduction to thermodynamics molar specific heat substance specific heat in calories temperature temperature and heat thermal conductivity thermal expansion and zeroth law of thermodynamics The

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Table of Contents Thermodynamics Multiple Choice Questions And Answers

1. Understanding the eBook Thermodynamics Multiple Choice Questions And Answers
 - The Rise of Digital Reading Thermodynamics Multiple Choice Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Thermodynamics Multiple Choice Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Thermodynamics Multiple Choice Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Thermodynamics Multiple Choice Questions And Answers
 - Personalized Recommendations
 - Thermodynamics Multiple Choice Questions And Answers User Reviews and Ratings

- Thermodynamics Multiple Choice Questions And Answers and Bestseller Lists
- 5. Accessing Thermodynamics Multiple Choice Questions And Answers Free and Paid eBooks
 - Thermodynamics Multiple Choice Questions And Answers Public Domain eBooks
 - Thermodynamics Multiple Choice Questions And Answers eBook Subscription Services
 - Thermodynamics Multiple Choice Questions And Answers Budget-Friendly Options
- 6. Navigating Thermodynamics Multiple Choice Questions And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Thermodynamics Multiple Choice Questions And Answers Compatibility with Devices
 - Thermodynamics Multiple Choice Questions And Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Thermodynamics Multiple Choice Questions And Answers
 - Highlighting and Note-Taking Thermodynamics Multiple Choice Questions And Answers
 - Interactive Elements Thermodynamics Multiple Choice Questions And Answers
- 8. Staying Engaged with Thermodynamics Multiple Choice Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Thermodynamics Multiple Choice Questions And Answers
- 9. Balancing eBooks and Physical Books Thermodynamics Multiple Choice Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Thermodynamics Multiple Choice Questions And Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Thermodynamics Multiple Choice Questions And Answers
 - Setting Reading Goals Thermodynamics Multiple Choice Questions And Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Thermodynamics Multiple Choice Questions And Answers
 - Fact-Checking eBook Content of Thermodynamics Multiple Choice Questions And Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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