### Solution Manual For Thermodynamics An Engineering Approach Chapter (1-18) 10th Edition

The properties of a substance is naturally dependent on

the behavior of the particles

The Macroscopic approach to thermodynamics does not require knowledge

of the behavior of individual molecules in a substance

The Macroscopic approach to thermodynamics is called

→ Classical Thermodynamics

The Microscopic approach to thermodynamics, a more elaborate approach, is based on the

average behavior of large groups of individual molecules

The Microscopic approach to thermodynamics is rather involved and known as

→ Statistical Thermodynamics

The Conservation of energy principle

Energy can change form from one to another, but it cannot be created or destroyed.

The change of energy content is expressed as

-> ΔE = E(in) - E(out)

The First Law of Thermodynamics is also known as

→ The Conservation of energy principle

The First Law of Thermodynamics asserts that

Energy is a thermodynamic property

The second law of thermodynamics asserts that

 Energy has quality and quantity and the processes occur in the direction of decreasing quality of energy

Quality of energy is the

→ Measurement of energy's ability to do work

# Thermodynamics An Engineering Approach Solution <a href="#">Chapter 15</a>

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