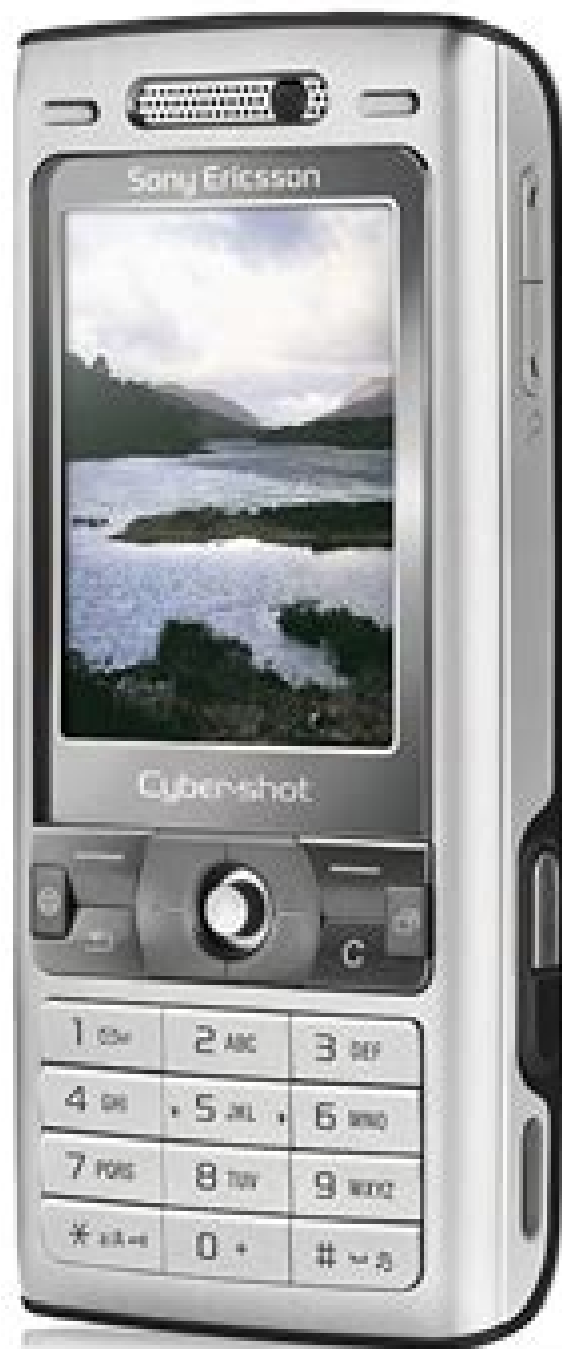




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web we will revisit this today and go on to see that if we use a special kind of thermochemical reaction called the standard enthalpy of formation we can calculate enthalpies of reactions without having to manipulate a series of individual thermochemical equations for each step

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web we can express the first law in many ways one of the more useful expressions is that the change in internal energy ΔE of a system in any process is equal to the heat q added to the system plus the work w done on the system by its surroundings $\Delta E = q + w$
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