

Experiment 1

1. Write the general equation for an equilibrium/phase transition and give the conditions of each. Example: melting

Experiment 2: Liquid and Vapor Equilibrium

For liquid-vapor equilibrium the following is given: a. Name the condition of the transition. b. Sketch the range of the transition.

- a. phase: _____ b. T / $^{\circ}\text{C}$: _____

a. _____ b. _____

a. _____ b. _____

Experiment 3: Solid and Vapor Equilibrium

- a. Sublimation/condensation of which substance? Name the process including conditions (a. T / $^{\circ}\text{C}$, b. p / bar) in the normal state.

- a. What phase transition process is shown?
Sublimation/condensation

- b. What substance is shown with the process?
Ice (water vapor)

- c. Sketch the range of the transition in the phase diagram.

- d. Sketch the process in the phase diagram.
Normal state: pressure, temperature

2. For a given substance, sketch the phase diagram. Give the normal state (a. T / $^{\circ}\text{C}$, b. p / bar) and the phase transition (a. T / $^{\circ}\text{C}$, b. p / bar) in the normal state.

- a. T / $^{\circ}\text{C}$: _____

- b. p / bar : _____

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