Worksheet 6.2 Word Equations

- Write the chemical equations and balance each of the following word equations.
 - Aluminum metal reacts with iron (ii) oxide powder to produce aluminum oxide solid and iron metal.

LAI +3FED -7 A1,0,+3FE

 Aluminum sulphate solution and calcium hydroxide solution produce a precipitate of aluminum hydroxide and solid calcium sulphate.

AL(50+) 3 +3 Ca(OH) = - 72AKOH) + 36aBOH

Ammonia gas (NH₂) plus oxygen gas yields nitrogen monoxide gas plus water vapour.

4NH, +10, -74NO + (H10

d) Calcium hydroxide solution and carbon dioxide gas yields solid calcium carbonate and liquid water.

(a(0H)) + CO2 -> (a(60))+ HIO

 Aqueous Iron (III) chloride and sodium carbonate solution yields aqueous sodium chloride and a precipitate of iron (III) carbonate.

2 Fellopes +) Markey ->6No (1 + Fc, (10))

f) Solid iron (III) axide and carbon monoxide gas yields iron metal and carbon dioxide gas.

Fe202 +260 -> 2Fe + 2602

Magnesium carbonate solution plus aqueous hydrochloric acid (HCI) yields magnesium chloride solution plus liquid water and carbon diexide gas.

My cos + 2HCI - Mycle + 400 + 600

 Silicon diaxide solid plus aqueous hydrofluoric acid (HF) yields solid silicon tetrafluoride plus liquid water.

5,020+44FGgs -> SiF46) +2H10

 Aqueous sodium hydroxide and carbon dioxide gas yields sodium carbonate solution and liquid water.

INOH + CO2 -7 NagCoz + HID

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