

COVALENT BONDING

Name _____

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons
all other atoms 8*



Note that hydrogen is content with 2, not 8, electrons.

electrons in the middle share

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

wanted available share

1. $\text{H} + \text{H} (\text{H}_2)$ W - 4 A - 2 S - 2	<i>wanted 2 + 2 available</i> $\text{H} \cdot \cdot \text{H}$ $\text{H} \cdot \quad \cdot \text{H}$	$\text{H} - \text{H}$ single bond = share 2 e ⁻
2. $\text{F} + \text{F} (\text{F}_2)$ W - 8 + 8 = 16 A - 7 + 7 = 14 S - 2	$\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot$	$\cdot \ddot{\text{F}} - \ddot{\text{F}} \cdot$ single bond 2 e ⁻
3. $\text{O} + \text{O} (\text{O}_2)$ W - 8 + 8 = 16 A - 6 + 6 = 12 S - 4	$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot$	$\cdot \ddot{\text{O}} = \ddot{\text{O}} \cdot$ 4 double bond
4. $\text{N} + \text{N} (\text{N}_2)$ W - 8 + 8 = 16 A - 5 + 5 = 10 S - 6	$\cdot \ddot{\text{N}} \cdot \cdot \ddot{\text{N}} \cdot$	$\cdot \ddot{\text{N}} \equiv \text{N} \cdot$
5. $\text{C} + \text{O} (\text{CO}_2)$ W - 8 + 8 + 8 = 24 A - 4 + 6 + 6 = 16 S - 8	$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{C}} \cdot \cdot \ddot{\text{O}} \cdot$	$\cdot \ddot{\text{O}} = \text{C} = \ddot{\text{O}} \cdot$
6. $\text{H} + \text{O} (\text{H}_2\text{O})$ W - 2 + 2 + 8 = 12 A - 1 + 1 + 6 = 8 S - 4	$\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}$	$\text{H} - \ddot{\text{O}} - \text{H}$

Ws 4 4 Bonding Ionic Covalent Answers

K Payea



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The Nature of Matter Gr. 5-8 , *Educart ICSE Class 10 One-shot Question Bank 2026 Chemistry (strictly for 2025-26 boards)* Sir Tarun Rupani, 2025-07-12 Fast track your Chemistry revision with this exam ready resource This One shot Question Bank by Sir Tarun Rupani is designed to help ICSE Class 10 students revise the complete Chemistry syllabus quickly and thoroughly It simplifies theory boosts numerical accuracy and ensures strong exam practice all aligned with the 2025 26 ICSE syllabus Key Features Strictly Based on ICSE 2025 26 Curriculum Complete chapter coverage including Periodic Table Chemical Bonding Acid Base Organic Chemistry and more One shot Format Each chapter includes concise concept notes chemical equations reactions and key diagrams for quick recall Complete Coverage of Question Types Includes objective short long answers equation based numerical and reasoning questions Chapterwise PYQs Included Practice with previous years ICSE board questions to understand trends and improve retention Solved Answers in ICSE Format Clear well structured solutions using proper units chemical symbols and balanced equations Smart Revision Focus Special tips to avoid common mistakes in writing reactions balancing equations and attempting numericals Why Choose This Book This Chemistry One shot by Sir Tarun Rupani is built for smart preparation whether you re revising at the last minute or practising throughout the term It helps you approach each question with clarity confidence and the precision needed to score high in the 2026 ICSE board exam *Prentice Hall Science Explorer* Michael J. Padilla, 2002 *Biology/science Materials* Carolina Biological Supply Company, 1991 **Academic Language/Literacy Strategies for Adolescents** Debra L. Cook Hirai, Irene Borrego, Emilio Garza, Carl Klock, 2013-02-01 Fast paced practical and innovative this text for pre service and in service teachers features clear easily accessible lessons and professional development activities to improve the delivery of academic language literacy education across the content areas in junior middle school and high school classrooms Numerous hands on tools and techniques demonstrate the effectiveness of content area instruction for students in a wide variety of school settings particularly English language learners struggling readers and other special populations of students Based on a strong professional development model the authors have been instrumental in designing Academic Language Literacy Strategies for Adolescents addresses motivation attributes of academic language vocabulary theory and practice reading skills development grammar and writing A wealth of charts graphs and lesson plans give clear examples of academic language literacy strategies in action The appendices a key component of the practical applications developed in the text include a glossary exemplary lessons that address key content areas and a Grammar Handbook In this era of increased accountability coupled with rapid demographic change and challenges to traditional curricula and pedagogical methods educators will find this book to be a great resource Holt Chemistry Holt Rinehart & Winston, 2003-01-24 **Exploring Earth and Space** Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter **Addison-Wesley Science**

Insights ,1996 Glencoe Science McGraw-Hill Staff,2001-08 The Physics and Chemistry of Solids ,1959 *Essential Biology* Neil A. Campbell,Jane B. Reece,Eric Jeffrey Simon,2007

Essential Biology is a brief non majors biology book that combines clear writing real world applications vivid art and powerful media to teach readers the important concepts of biology and give them an appreciation for how biology relates to their everyday lives In the Second Edition best selling authors Neil Campbell and Jane Reece are joined by Eric Simon who uses his experience teaching non majors biology to keep the book both accessible and up to date To help readers become informed citizens the new edition features even more human applications and up to date information on important issues like DNA technology cloning and global warming

KEY TOPICS

The book covers four major biological topics cells genetics evolution diversity and ecology and uses evolution as an overarching theme to tie all 20 chapters together For college instructors students or anyone interested in biology

Advances in Semiconductor Science ,1959 *Proceedings* ,1959 **The Journal of Physics and Chemistry of Solids** ,1959 *Earth Science* Samuel N. Namowitz,Nancy E. Spaulding,1989 **Chemistry** John S. Phillips,Cheryl Wistrom,2000

Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 **Manual of the Public Examinations Board** University of Adelaide. Public Examinations Board,1965

The Manuals include information on syllabus regulations copies of examination papers and notes by examiners They also include pass lists **Holt Science and Technology** Holt, Rinehart and Winston Staff,2001 **Holt Biology** Holt Rinehart & Winston,2003-08

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Ws 4 4 Bonding Ionic Covalent Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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