

Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl₂", as shown.

	zinc	iron (II)	iron (III)	gallium	silver	lead (IV)
chloride	ZnCl ₂	FeCl ₂	FeCl ₃	GaCl ₃	AgCl	PbCl ₄
acetate	Zn(C ₂ H ₃ O ₂) ₂	Fe(C ₂ H ₃ O ₂) ₂	Fe(C ₂ H ₃ O ₂) ₃	Ga(C ₂ H ₃ O ₂) ₃	Ag(C ₂ H ₃ O ₂)	Pb(C ₂ H ₃ O ₂) ₄
nitrate	Zn(NO ₃) ₂	Fe(NO ₃) ₂	Fe(NO ₃) ₃	Ga(NO ₃) ₃	AgNO ₃	Pb(NO ₃) ₄
oxide	ZnO	FeO	Fe ₂ O ₃	Ga ₂ O ₃	Ag ₂ O	PbO ₂
nitride	Zn ₃ N ₂	Fe ₃ N ₂	FeN	GaN	Ag ₃ N	Pb ₃ N ₄
sulfate	ZnSO ₄	FeSO ₄	Fe ₂ (SO ₄) ₃	Ga ₂ (SO ₄) ₃	Ag ₂ SO ₄	Pb(SO ₄) ₂

Write the formulas for the following compounds:

- copper (II) chloride **CuCl₂**
- lithium acetate **LiC₂H₃O₂**
- vanadium (III) selenide **V₂Se₃**
- manganese (IV) nitride **Mn₃N₄**
- beryllium oxide **BeO**
- sodium sulfate **Na₂SO₄**
- aluminum arsenide **AlAs**
- potassium permanganate **KMnO₄**
- chromium (VI) cyanide **Cr(CN)₆**
- tin (II) sulfite **SnSO₃**
- vanadium (V) fluoride **VF₅**
- ammonium nitrate **NH₄NO₃**

Writing Chemical Word Equations

**Raymond B. Brownlee, Robert W.
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