

WRITING FORMULAS FROM NAMES

Name _____

Write the formulas for the following compounds.

1. carbon monoxide _____
2. sodium chloride _____
3. carbon tetrachloride _____
4. magnesium bromide _____
5. aluminum iodide _____
6. hydrogen hydroxide _____
7. iron (II) fluoride _____
8. carbon dioxide _____
9. sodium carbonate _____
10. ammonium sulfide _____
11. iron (II) oxide _____
12. iron (III) oxide _____
13. magnesium sulfate _____
14. sodium phosphate _____
15. dinitrogen pentoxide _____
16. phosphorus trichloride _____
17. aluminum sulfite _____
18. copper (I) carbonate _____
19. potassium hydrogen carbonate _____
20. sulfur trioxide _____

Writing Formulas Physical Science If8767

Robert J. Lefkowitz, Edmund Eisenberg



Writing Formulas Physical Science If8767:

Compounds and Formulas Lib Foster, Leigh Nix, Auburn University, 1987 **Writing Chemical Formulas** Robert J. Lefkowitz, Edmund Eisenberg, 1983 Chemical formulas and names Virginia P. Powell, 1965 *Formula Writing and Nomenclature Explained (Chemistry Review)* E Staff, Learn and review on the go Use Quick Review Science Study Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students **Formulas, Facts, and Constants** H. J. Fischbeck, K. H. Fischbeck, 2012-12-06 The material presented in this book has been compiled for the convenience of the reader The aim of the book is to provide a handy source of formulas conversion factors and constants for everyday use The formulas and tables are amended by examples in all of those cases where their use is not self explanatory The material has been selected to be helpful whenever it is inconvenient or not possible to consult tables available at the library Section 1 provides the fundamental tools of mathematics needed in all areas of the physical sciences Section 2 summarizes the SI system lists conversion factors and provides precise values of fundamental constants Sections 3 and 4 review the basic terms of spectroscopy atomic structure and wave mechanics These sections serve as a guide to the interpretation of modern literature Section 5 is a resource for work in the laboratory Data and formulas are given to be of assistance in the use of frequently encountered equipment such as vacuum systems and electronic devices Material constants and other data are listed for information and as an aid for estimates or problem solving The assistance of the Springer Verlag during the various stages of the development of this book is gratefully acknowledged The authors like to thank Dr F L Boschke for his many helpful suggestions Helmut J Fischbeck Kurt H Fischbeck Contents Basic mathematical facts and figures **Formulas, Facts, and Constants for Students and Professionals in Engineering, Chemistry, and Physics** Helmut J. Fischbeck, Kurt Fischbeck, 1982 **Formulas & Equation Writing**, 2006 **All of Physics (Almost) in 15 Equations** Bruno Mansoulié, 2018

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