Balancing Equations Worksheet

Using your newly-found equation balancing knowledge, balance the following equations:

1) ____ Br2 + ___ LIF -> ___ LIBr + ___ F2

2) ____H₃PO₄ + ____Fe₃(PO₄)₂ → ____H₂O + ____Fe₃(PO₄)₂

4) ____ Ni(OH)3 → ____ Ni₂O₃ + ____ H₂O

5) ____ K₂SO₃ + ____ Mn(OH)₂ → ____ KOH + ___ MnSO₃

6) ____NaOH + ___H2SO4 -> ___H2O + ___Na2SO4

7) ____ Li + ____ Pb(OH)2 → ____ Pb + ____ LiOH

8) ____ C₄H₈ + ____ O₂ \rightarrow ____ CO₂ + ____ H₂O

9) ___ Ga(OH)3 + ___ KF → ___ KOH + ___ GaF3

10) ____ V + ___ ZnBr₂ → ___ VBr₃ + ___ Zn

11) ____ As₂O₅ + ____ H₂O → ____ H₃AsO₄

12) ____ NH₃ + ____ O₂ → ____ NO + ____ H₂O

13) ___ $C_3H_8 + __ O_2 \rightarrow __ CO_2 + __ H_2O$

14) ____ NaClO₃ → ____ NaCl + ____ O₂

15) ___ Ca + ___ O₂ → ___ CaO

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Stanford Linear Accelerator Center. Computation Group, S. N. Au

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Compounds and Formulas Lib Foster, Leigh Nix, Auburn University, 1987

Chemical Formulae and Equations, 1973

Balancing and Naming Simple Binary Equations for Beginning Chemistry Students Harold Querner, 1994

Evaluates a drill and practice computer program and teaching unit for the teaching of balancing chemical equations in a beginning chemistry class

Basic Equations of Engineering Science William F. Hughes, Eber W. Gaylord, 1954

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