

Total No. of Questions : 6]

SEAT No. :

P462

[Total No. of Pages : 3

[5115] - 205
S.Y. B.Sc. (Semester - II)
CHEMISTRY
CH - 221: Physical and Analytical Chemistry
(2013 Pattern)

Time : 2 Hours]

[Max. Marks : 40

Instructions to the candidates:

- 1) All questions are compulsory.
- 2) Neat diagram must be drawn wherever necessary.
- 3) Figures to the right side indicate full marks.
- 4) Use of calculator is allowed.
- 5) Answer to both sections should be written in separate answer sheet.

SECTION - I
[Physical Chemistry]

Q1) Answer the following. [5]

- a) State the criteria of equilibrium in terms of ΔF .
- b) Write the Gibbs - Helmholtz equation.
- c) Draw the T-N diagram of solution with negative deviation.
- d) Define the term: Lower critical solution temperature.
- e) Define the term: Normality.

Q2) a) Attempt any two of the following. [6]

- i) Derive the Clapeyron equation.
- ii) State and explain Henry's law. Give its applications.
- iii) Explain with diagram the mutual solubility behaviour of phenol - water system at various temperatures.

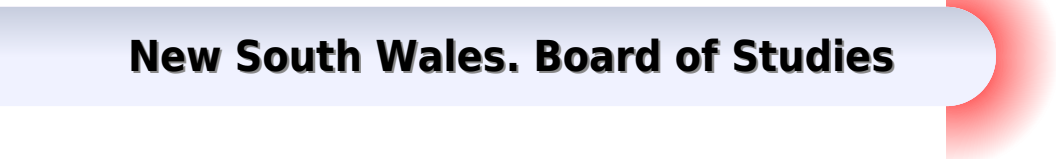
b) Answer any two of the following. [4]

- i) State Raoult's law and give its equation.
- ii) Explain the physical significance of ΔF .
- iii) How will you calculate graphically the heat of vaporisation of liquid from Clausius - Clapeyron equation.

P.T.O.

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Chemistry 5 Syllabus Spencer L. Seager, Stoker, H. Stephen, 196? **Syllabus for Chemistry Curriculum** Development Council (Hong Kong, China), 1991 **General Chemistry Syllabus** Orin Fletcher Stafford, 1931 Practical Chemistry J. Howard, 1873 **General Chemistry Syllabus** Orin Fletcher Stafford, 1931 *Chemistry*, 1998 Syllabus for Chemistry University of Chicago. Department of Chemistry, 1931 **CHEMISTRY PHASE 5** Dronadula Basavaiah, 2020-02-20 This book is part of 7 books which covers whole chemistry for the Board as well competitive exams We have divided total chemistry syllabus in to 7 books each one will come in every phase All of these books are designed to keep in mind the requirements of CBSE board as well IIT entrance exam syllabus This Book consists of Nomoreclass concepts and previous IIT questions **Chemistry Syllabus** New South Wales. Board of Studies, 1990 **Chemistry for Degree Students B.Sc. Semester - IV (As per CBCS)** Madan R.L., 2017 This textbook has been designed to meet the needs of B Sc Fourth Semester students of Chemistry as per the UGC Choice Based Credit System CBCS With its traditional approach to the subject this textbook lucidly explains principles of chemistry Important topics such as transition elements coordination chemistry crystal field theory kinetic theory of gases liquids solids and chemical kinetics are aptly discussed to give an overview of inorganic and physical chemistry Laboratory work has also been included to help students achieve solid conceptual understanding and learn experimental procedures **America's First Chemistry Syllabus-and-course for Girls** Wyndham D. Miles, Harold J. Abrahams, 1958 **Chemistry** Ying-wai Chan, 1981 Chemical Syllabus John Vaughan, 1799 *Chemistry Syllabus* Pearson Custom Publishing, 1994-05-01 *Report on the examination in chemistry*, 1993 **Chemistry Senior Syllabus 2007** Queensland Studies Authority, 2007 **Practical Chemistry** John Howard, 1872 *Chemistry Senior Syllabus*, 1995 **Chemistry syllabus** Caribbean Examination Council, 1987 **B.Ed Chemistry Programme**, 1993

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