C-100 Practice Multiple Choice Question Set 2

1. When the following reaction equation is properly balanced, the number of moles of O2 will be ...

- (a) 1.5 moles
- (b) 3.5 moles
- (c) 3 modes
- (d) 5 moles
- 2. When the following reaction equation is properly balanced, the number of moles of O2 will be...

C₆H₁₄ + C₂ ------ CO₂ + H₂O

- (a) 1.5 moles
- (b) 13 moles
- (c) 19 moles
- (d) 38 moles
- How many moles of CO₂ are present in 220 mg?
 - (a) 5 moles
 - (b) 0.005 mole
 - (c) 5000 moles
 - (d) 10 moles
- 4. If the temperature of a reaction is increased by 20°C, the reaction rate will be...
 - (a) 2 times as fast.
 - (b) 4 times as fast.
 - (c) 20 times as fast.
 - (d) unchanged because the reaction rate is not dependent on the temperature.
- One mole of (NH₄)₂HPO₄ contains _?_ moles of hydrogen atoms.
 - (a) 1
 - (b) 5
 - (c) 6
 - (d) 9
- 6. Which of the following compounds should have the highest boiling point?
 - (a) NH₃
 - (b) CH₄
 - (c) H₂O
 - (d) HF
- 7. The following reaction coordinate diagram represents...



- an endothermic reaction.
- (b) an exothermic reaction.

Static Equilibrium Multiple Choice Questions

Stephen Grounds, Steven Grounds, Edwin Kirby

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