Answers to Chapter 3 Study Questions

1. a)
$$3(12.0) + 8(1.01) + 3(16.0) = 92.1 \text{ g/mole}$$
 b) 92.1 g c) $6.02 \times 10^{23} \text{ molecules}$ d) $0.217 \text{ moles} \times \frac{92.1 \text{ g}}{1 \text{ mole}} = 20.0 \text{ g}$

b) 1 mole NH₃ x
$$\frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole NH}_3}$$
 x $\frac{4 \text{ atoms}}{1 \text{ molecule}}$ = 2.41 x 10²⁴ atoms

c)
$$3.40 \text{ g NH}_3$$
 $x \frac{1 \text{ mole NH}_3}{17.0 \text{ g NH}_3} x \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole NH}_3} x \frac{4 \text{ atoms}}{1 \text{ molecule}} = 4.82 \times 10^{23} \text{ atoms}$

$$40.7 \text{ g C} \times \frac{1 \, mole \, C}{12.0 \, \text{g C}} = 3.39 \, \text{moles C}$$
 $3.39/3.39 = 1 \times 2 = 2$
 $5.1 \, \text{g H} \times \frac{1 \, mole \, H}{1.0 \, \text{g H}} = 5.1 \, \text{moles H}$ $5.1/3.39 = 1.5 \times 2 = 3$
 $54.2 \, \text{g O} \times \frac{1 \, mole \, O}{16.0 \, \text{g O}} = 3.39 \, \text{moles O}$ $3.39/3.39 = 1 \times 2 = 2$

b) Molar mass of
$$C_2H_3O_2 = 2(12.0) + 3(1.0) + 2(16.0) = 59.0$$
 g/mole $118/59.0 = 2 \rightarrow$ molecular formula = $C_4H_6O_4$

b) %
$$Mg = 7.20/25.0 = 28.8\%$$
 Mg; %C = 3.55/25.0 = 14.2 % C; % O = 14.25/25.0 = 57.0% O. (You should get the same result using molar mass and atomic masses.)

c) 13.9 g compound ×
$$\frac{288 g Mg}{100 g compound}$$
 = 4.00 g Mg

Stars Suite Chemistry Answers

Justin P. Lomont, Ian C. Stewart

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The Handy Chemistry Answer Book Justin P. Lomont, Ian C. Stewart, 2013-10-01 Don t be mixed up about chemistry Simplify the complex chemical reactions that take place everywhere in our lives with this engaging easy to follow question and answer guide Where would we be without atoms and compounds Gas liquids solids and plasma Acids and bases Bonds and reactions Matter and energy The Handy Chemistry Answer Book covers the building blocks of life and the universe The secret life of atoms how polar bears aren t actually white why oil and water don t mix and much much more are revealed and explained This informative guide covers the basics of chemistry history atomic structures chemical bonds and reactions organic and inorganic chemistry to more advanced material nuclear chemistry biochemistry physical and theoretical chemistry by answering nearly 1 000 common chemistry questions including What causes lightning How does photosynthesis work What are hard and soft Lewis acids and bases What makes a fabric waterproof What are the twelve principles of green chemistry When did alchemists finally abandon trying to make gold What is Le Chatelier's principle What do the different octane ratings mean at the gas pump What is genetic engineering Why is calcium important for strong bones What is the 18 electron rule Why does chocolate turn white as it ages Chemical reactions that rule the world their properties structure composition behavior and history are tackled and explained in plain English in The Handy Chemistry Answer Book With many photos illustrations a few formulas molecular diagrams and other graphics this fun fact filled tome is richly illustrated A history of chemistry timeline appendices on Nobel Prize in Chemistry winners a bibliography further reading section glossary of terms a table of physical constants a table of conversion factors and extensive index add to its usefulness 15 PGT Chemistry Test Papers EMRS Mocktime Publication, EMRS Exam Teachers PGT Chemistry Test Papers 15 Practice Papers Tier 1 Eklavya Model Residential Schools as per Official Exam Pattern and Syllabus **Boston Journal of Chemistry and** Pharmacy, 1866 The Last Stargazers Emily Levesque, 2020-08-04 The story of the people who see beyond the stars an astronomy book for adults still spellbound by the night sky Embark on a captivating cosmic journey with The Last Stargazers This enthralling book takes you on an awe inspiring exploration of the night sky offering a unique perspective on the vast celestial wonders that have fascinated humanity for millennia Written by astrophysicist Dr Emily Levesque The Last Stargazers combines scientific expertise with captivating storytelling making it the perfect companion for both astronomy enthusiasts and curious minds Dr Levesque's passion for the stars shines through as she shares her personal experiences and encounters while working at some of the world's most renowned observatories. Delve into the fascinating world of astronomy as you uncover the secrets of distant galaxies supernovae and elusive celestial phenomena Discover Inspiring narratives Dr Levesque's engaging storytelling transports readers to the front lines of astronomical research providing a behind the scenes glimpse into the life of a modern day stargazer Cutting edge research Stay up to date with the latest scientific breakthroughs and advancements in the field of astronomy as Dr Levesque shares her firsthand experiences and encounters Accessible

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his story but it s also a universal one an honest funny sometimes raw and often inappropriate glimpse into the mind of a young man dealing with a life threatening illness in the only way he knows how by laughing in its face **Popular Science**News ,1866 Marine Engineering Log ,1908 The Eclectic Magazine John Holmes Agnew,Walter Hilliard Bidwell,1850

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