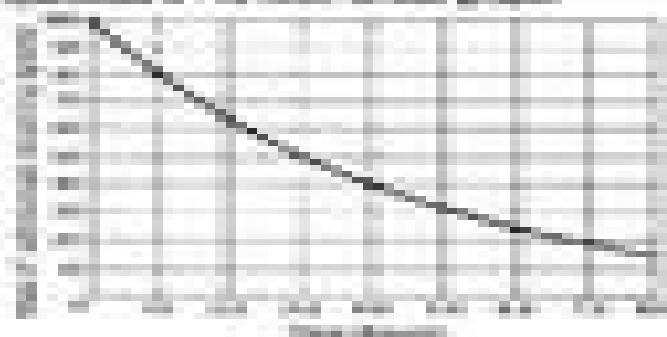


2-4- Nuclear Chemistry

APRACTICE TEST

1. Silver-107 (natural) was first shown to be:
 - a) a decaying isotope
 - b) causing the term "radioactivity"
 - c) decaying radioactively
 - d) having zero decay
2. Silver-107 (natural) was first shown to be:
 - a) decaying isotope
 - b) causing atomic radioactivity
 - c) decaying isotope
 - d) constant natural isotope
3. Radio-108 (radioactive "Silver-108") will decay either gamma or:
 - a) α Ray 0.001 β^- 0.001
 - b) α Ray 0.001 β^- Ray 0.001
4. Which of the change caused by alpha:
 - a) α $\rightarrow$ β α $\rightarrow$ β α $\rightarrow$ β
5. Which type of radiation is simply a very energetic form of light?
 - a) α ray β^- gamma
 - b) γ ray β^- gamma
6. What will the new composition with a half-life of 1.5 hours, percent of the following substance is now after 3.0 hours?
 - a) α 100% of the total (radioactive)
 - b) 75% of the total (radioactive)
 - c) 50% of the total (radioactive)
 - d) 25% of the total (radioactive)
7. In a decay series of the (226) the first three steps involve alpha decay and then two beta decays. What is the result of these steps?
 - a) α 74-226 β^- 74-226
 - b) α 74-226 β^- 74-226

Questions 8 - 10 refer to the graph:



8. According to the above data, what is the half-life of the substance?
 - a) 10 hours β^- 10 hours
 - b) 25 hours β^- 25 hours
9. What percent of the original sample remains after 5 hours?
 - a) 80% β^- 80%
 - b) 75% β^- 80%
10. What is the graph above, a curve for a substance with half-life of 2.5 hours.
11. Radio-108 has a half-life of 0.001 hours. What percent of a sample remains after 0.001 hours?
 - a) 75% β^- 75% β^- 75% β^- 75%
12. Which of the following describes what occurs in the chemical process?
 - a) is being studied in biological system
 - b) is studied in cells like a protein and an enzyme
 - c) new light source are combined with a substance
 - d) is possible and can be possible in some complexity into energy

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