

Chem 210 Final exam study guide, Questions and Correct Answers, 2024. With complete solution.

What is ground state electron configuration of Co?

[Ar]3d⁹

[Ar]4s¹ 3d⁸

[Ar]4s² 3d⁷

[Ar]4s² 4p⁶ 4d¹

[Ar]4s² 3d⁷

molybdenum has an anomalous electron configuration. using the shorthand notation for electron configurations, write the electron configuration

A) [kr]5s 4d⁶

B) [kr]5s 4d⁰ 5p⁶

C) [kr]5s¹ 4d⁵

D) [kr]5s² 4d⁴

C) [kr]5s¹ 4d⁵

the symbol [Kr] represents

a) 4s² 4p⁶

b) 1s² 2s² 2p⁶ 3s⁶ 4s² 4p⁶

c) 1s² 2s² 2p⁶ 3s⁶ 4s² 3d¹⁰ 4p⁶

D) 1s² 2s² 2p⁶ 3s⁶ 4s² 3d¹⁰ 4p⁶ 4d¹⁰

c) 1s² 2s² 2p⁶ 3s⁶ 4s² 3d¹⁰ 4p⁶

What is the wavelength of UV light in the balmer series when m=2 and n=7?

a) 0.00252 nm

b) 397 nm

c) 337 nm

d) 0.00297 nm

b) 397 nm

How many electrons can a single orbital hold?

A) 2n

B) 2

C) 2l+1

D) 8

B) 2

What are the possible values of l and m_l for an electron in a 5d orbital?

A) l=0 and m_l=0

B) l=3 and M_l=-2,-1,0,+1, or +2

c) l=1 and M_l=-1,0,+1

D) l=2 and M_l=-2,-1,0,+1, +2

D) l=2 and M_l=-2,-1,0,+1, +2

What is the the de Broglie wavelength of an electron (m=9.11e-31 kg) moving at a velocity of 3.0e7m/s?

a) 3.9e-12

b) 2.4e-11

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