



Vaal University Of Technology Prospectus 2015

S Marginson



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consider the following mechanism
 $\text{O}_3 \xrightarrow{\text{fast}} \text{O}_2 + \text{O}$
 $\text{O} + \text{O}_2 \xrightarrow{\text{slow}} \text{O}_3$
a) write the overall balanced chemical equation
b) identify any intermediates within the mechanism

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 $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
g the reaction between Mg and HCl is represented by the equation above in a kinetics experiment
a) 0.080 g sample of Mg ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M HCl at 25 °C

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reactions conducted in an industrial facility mix compounds together heating and stirring them for a

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