

Name: _____

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Chemistry
Assessment #20: Water & Aqueous Systems PRACTICE

For each of the following questions or statements, select the most appropriate response and write its letter on the answer line:

- _____ 1. Which of the following properties of water can NOT be attributed to hydrogen bonding?
- A. large molar mass
 - B. high boiling point
 - C. large heat capacity
 - D. less dense solid phase
- _____ 2. Each of the following terms can correctly be used to describe water molecules EXCEPT:
- A. bent dipole
 - B. dilute
 - C. polar
 - D. triatomic
- _____ 3. The phrase "like dissolves like" refers to the fact that:
- A. gases can only dissolve other gases
 - B. polar solvents dissolve polar solutes and nonpolar solvents dissolve nonpolar solutes
 - C. solvents can only dissolve solutes of similar molar mass
 - D. polar solvents dissolve nonpolar solutes and vice versa
- _____ 4. The property responsible for the "beading up" of water is:
- A. viscosity
 - B. vapor pressure
 - C. surface tension
 - D. density
- _____ 5. The process of solute particles being surrounded by solvent particles is known as:
- A. salutation
 - B. agglomeration
 - C. solvation
 - D. dehydration
- _____ 6. Which of the following is not a colloid?
- A. air
 - B. fog
 - C. smoke
 - D. whipped cream

Water And Aqueous Assessment Chemistry Answers

Carolyn Chapman



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