

6 Standardized Test Practice

SCORE _____

(Chapters 1–6)

Part 1: Multiple Choice**Instructions:** Fill in the appropriate circle for the best answer.

1. Find the coordinates of
- X
- if
- $H(0.5, 5)$
- is the midpoint of
- $\overline{UX}$
- with
- $U(15, 21)$
- .

(Lesson 1-3)

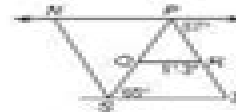
A $(-14, -11)$ **C** $(0, 0)$ **B** $(7.75, 22.5)$ **D** $(15.5, -5)$ 1. **(A)** **(B)** **(C)** **(D)**

2. Which of the following are possible measures for vertical angles
- G
- and
- H
- ?

(Lesson 2-8)

F $m\angle G = 125$ and $m\angle H = 55$ **G** $m\angle G = 125$ and $m\angle H = 125$ **H** $m\angle G = 55$ and $m\angle H = 45$ **J** $m\angle G = 55$ and $m\angle H = 152.5$ 2. **(F)** **(G)** **(H)** **(J)**

3. Determine which lines are parallel. (Lesson 3-5)

A $\overline{NS} \parallel \overline{PT}$ **C** $\overline{QR} \parallel \overline{ST}$ **B** $\overline{NP} \parallel \overline{ST}$ **D** $\overline{NP} \parallel \overline{QR}$ 3. **(A)** **(B)** **(C)** **(D)**

4. Find the coordinates of
- B
- , the midpoint of
- $\overline{AC}$
- , if
- $A(2a, b)$
- and
- $C(0, 2b)$
- .

(Lesson 4-8)

F $(2a, 2b)$ **G** (a, b) **H** $(a, \frac{3}{2}b)$ **J** $(\frac{3}{2}a, b)$ 4. **(F)** **(G)** **(H)** **(J)**

5. If
- $\overline{RV}$
- is an angle bisector, find
- $m\angle UVT$
- . (Lesson 5-1)

A 10**C** 68**B** 34**D** 1365. **(A)** **(B)** **(C)** **(D)**

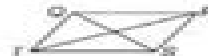
6. Find the slope of the line that passes through points
- $A(-7, 14)$
- and
- $B(5, -2)$
- .

(Lesson 3-3)

F $-\frac{2}{3}$ **G** $-\frac{3}{4}$ **H** $\frac{3}{4}$ **J** $\frac{4}{3}$ 6. **(F)** **(G)** **(H)** **(J)**

7. Which statement ensures that quadrilateral
- $QRST$
- is a parallelogram?

(Lesson 6-3)

A $\angle Q \cong \angle S$ **C** $\overline{QT} \parallel \overline{RS}$ **B** $\overline{QR} \cong \overline{TS}$ and $\overline{QR} \parallel \overline{TS}$ **D** $m\angle Q + m\angle S = 180$ 7. **(A)** **(B)** **(C)** **(D)**

Standardized Test Prep Chemistry Ch 16

Carl A. Burtis, David E. Bruns



Standardized Test Prep Chemistry Ch 16:

Holt Chemistry Ralph Thomas Myers, 2004 *Prentice Hall Physical Science Concepts in Action Program Planner*
National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **SAT Subject Test: Chemistry with Online Tests** Joseph A. Mascetta, Mark Kernion, 2018-09-01 Always study with the most up to date prep Look for SAT Subject Test Chemistry ISBN 9781506263120 on sale December 01 2020 Publisher s Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entitles included with the product **Essentials of Physical Chemistry** Don Shillady, 2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author s thirty years of teaching Essentials of Physical Chemistry merges coverage of calculus with chemist *Essentials Of Medical Laboratory Practice* Constance L Lieseke, Elizabeth A Zeibig, 2012-01-31 Inside you ll find a wealth of information on important laboratory terminology and the procedures you ll need to perform to become an effective member of a physician s office team Coverage of the advanced procedures performed outside of the physician s office explains what happens to the samples you send out There s also information on CLIA and other government regulations and how they affect each procedure **SAT Subject Test Chemistry** Joseph A. Mascetta, Mark Kernion, 2020-12-01 The updated edition of Barron s SAT Subject Test Chemistry includes A full length diagnostic test with explained answers Four practice tests that reflect the actual SAT Subject Test Chemistry All questions answered and explained Detailed reviews covering all test topics Appendixes which include the Periodic Table important equation constant and data tables and a glossary of chemistry terms Both teachers and test taking students have praised earlier editions of this manual for its wealth of well organized detail Subject reviewed include the basics matter energy scientific method and measurements atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory **ONLINE PRACTICE TESTS** Students who purchase this book or package will also get access to two additional full length online SAT Chemistry subject tests with all questions answered and explained **ACT Prep 2026/2027 For Dummies** Lisa Zimmer Hatch, Scott A. Hatch, 2025-08-18 The trusted study guide updated for the latest changes to the ACT exam ACT Prep 2026 2027 For Dummies is your go to resource for conquering the ACT and embarking on your post high school journey with confidence A high score on this nationally recognized college entrance exam can boost your chances of

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