ACS Final Exam Practice 1.

The exam is 70 questions in 110 minutes. You may use the equation sheet and periodic table but no other materials

Section 1 Foundations

PQ-1. The speed of light is 3.0 × 10 m·s 1. What is this speed in nm·ms 1?

(A) 3.0×10⁻¹ nm-ms⁻¹

(B) 3.0×10² nm·ms ¹

(C) 3.0×10⁻⁴ nm-ms⁻¹ (D) 3.0×10²⁰ nm-ms⁻¹

Mass of beaker and water

PO-4. What is the correctly reported mass of water based on the data in the salde?

(A) 1.3 g

1.30 g

Mass of bealer only

(C) 1,298 g

(D) 1.2980 g

29.62 亩

28.3220 m

PQ-5. What is the name of Ti₂(PO₄)₄?

titanium phosphate

citanium(IV) phosphate

(B) titanium(III) phosphate

(D) titanium tetruphosphate

PQ-6. The formula of strontium bexafluorosilicate is SrSil's. What is the formula of aluminum hexafluorosilicate?

(A) AlSiFa

Alp(SiFa)s

(C) AbSiFe

(D) Ab(SiFa):

PO-8. What is the correct name for N-O-7.

(A) nitrogen(II) oxide

(C) dinitrogen oxide

mitrogen(V) oxide

dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of 12.0 g-cm⁻³, what is the mass of a nanoparticle with a volume of 1.84×10-21 cm3?

2.21×10-20 g

(B) 1.63×10⁻³⁰ g

(C) 6.52×10⁻²¹ at

(D) 1.53×10⁻²² g

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M Woodhall

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Journal of Agricultural and Food Chemistry, 1954 U.S. Government Research Reports .1961 Introduction to Pharmaceutical Analytical Chemistry Stig Pedersen-Bjergaard, Bente Gammelgaard, Trine G. Halvorsen, 2019-04-29 The definitive textbook on the chemical analysis of pharmaceutical drugs fully revised and updated Introduction to Pharmaceutical Analytical Chemistry enables students to gain fundamental knowledge of the vital concepts techniques and applications of the chemical analysis of pharmaceutical ingredients final pharmaceutical products and drug substances in biological fluids A unique emphasis on pharmaceutical laboratory practices such as sample preparation and separation techniques provides an efficient and practical educational framework for undergraduate studies in areas such as pharmaceutical sciences analytical chemistry and forensic analysis Suitable for foundational courses this essential undergraduate text introduces the common analytical methods used in quantitative and qualitative chemical analysis of pharmaceuticals This extensively revised second edition includes a new chapter on chemical analysis of biopharmaceuticals which includes discussions on identification purity testing and assay of peptide and protein based formulations Also new to this edition are improved colour illustrations and tables a streamlined chapter structure and text revised for increased clarity and comprehension Introduces the fundamental concepts of pharmaceutical analytical chemistry and statistics Presents a systematic investigation of pharmaceutical applications absent from other textbooks on the subject Examines various analytical techniques commonly used in pharmaceutical laboratories Provides practice problems up to date practical examples and detailed illustrations Includes updated content aligned with the current European and United States Pharmacopeia regulations and guidelines Covering the analytical techniques and concepts necessary for pharmaceutical analytical chemistry Introduction to Pharmaceutical Analytical Chemistry is ideally suited for students of chemical and pharmaceutical sciences as well as analytical chemists transitioning into the field of pharmaceutical analytical chemistry

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