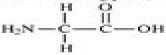
Ch 14 Practice MC problems

- Which of the following molecules contains a nitrogen atom that is sp² hybridized?
 - A) NH₂
 - B) NO₃:
 - C) N₂
 - D) HCN
 - E) C₀N₀
- 2. Consider the structure of glycine, the simplest amino acid:



What is the total number of π bonds in the molecule?

- A0 = 0
- B) 172.
- C) 1
- D) 2
- E) More information is needed.
- What is the hybridization of I in the molecule ICh?
 - A) sc
 - B) sp²
 - C) sp³
 - D) dsp³
 - E) d'sp³
- 4. What is the hybridization of S in the molecule H₂S?
 - A3 39
 - B) sp²
 - C) sp⁸
 - D) dsp³
 - E) d'sp'
- What is the hybridization of C in the ion CN?
 - A) sp
 - B) sp²
 - C) sp^N
 - D) dsp^b
 - E) d'sp'
- Atoms that are sp¹ hybridized form ____ pi bond(s).
 - A(3) = 0
 - B) 1
 - C) 2
 - D) 3
 - E) 4
- What is the hybridization of the central atom in SF₆?
 - A) sp
 - B) sp²
 - C) sp³
 - D) dsp³
 - E) d^2sp^3

Standardized Test Practice Chapter 14 Chemistry Answers

Arshad Iqbal

Standardized Test Practice Chapter 14 Chemistry Answers:

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