Standard Heats of Formation

Some Heats of Formation, ΔH_f^* (kJ/mol)

| CO(g) | -111 | $C_2H_2(g)$ | 227 | Ag+(aq) | 106 |
|-------------|------|-----------------------|------|-------------------------------------|-------|
| $CO_2(g)$ | -394 | $C_2H_4(g)$ | 52 | Na ⁺ (aq) | -240 |
| $H_2O(l)$ | -286 | $C_2H_6(g)$ | -85 | NO ₃ -(aq) | -207 |
| $NH_3(g)$ | -46 | CH ₃ OH(g) | -201 | Cl ⁻ (aq) | -167 |
| $N_2H_4(g)$ | 95.4 | $C_2H_5OH(g)$ | -235 | AgCl(s) | -127 |
| HCl(g) | -92 | $C_6H_6(l)$ | 49 | Na ₂ CO ₃ (s) | -1131 |

Standard Heat Of Formation Answer Key

United States. National Bureau of Standards

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