

Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl₂", as shown.

	<i>zinc</i>	<i>iron (II)</i>	<i>iron (III)</i>	<i>gallium</i>	<i>silver</i>	<i>lead (IV)</i>
<i>chloride</i>	ZnCl₂	FeCl₂	FeCl₃	GaCl₃	AgCl	PbCl₄
<i>acetate</i>	Zn(C₂H₃O₂)₂	Fe(C₂H₃O₂)₂	Fe(C₂H₃O₂)₃	Ga(C₂H₃O₂)₃	Ag₂C₂H₃O₂	Pb(C₂H₃O₂)₄
<i>nitrate</i>	Zn(NO₃)₂	Fe(NO₃)₂	Fe(NO₃)₃	Ga(NO₃)₃	AgNO₃	Pb(NO₃)₄
<i>oxide</i>	ZnO	FeO	Fe₂O₃	Ga₂O₃	Ag₂O	PbO₂
<i>nitride</i>	Zn₃N₂	Fe₃N₂	FeN	GaN	Ag₃N	Pb₃N₄
<i>sulfate</i>	ZnSO₄	FeSO₄	Fe₂(SO₄)₃	Ga₂(SO₄)₃	Ag₂SO₄	Pb(SO₄)₂

Write the formulas for the following compounds:

- 1) copper (II) chloride **CuCl₂**
- 2) lithium acetate **LiC₂H₃O₂**
- 3) vanadium (III) selenide **V₂Se₃**
- 4) manganese (IV) nitride **Mn₃N₄**
- 5) beryllium oxide **BeO**
- 6) sodium sulfate **Na₂SO₄**
- 7) aluminum arsenide **AlAs**
- 8) potassium permanganate **KMnO₄**
- 9) chromium (VI) cyanide **Cr(CN)₆**
- 10) tin (II) sulfite **SnSO₃**
- 11) vanadium (V) fluoride **VF₅**
- 12) ammonium nitrate **NH₄NO₃**

Writing Chemical Formula Answers

Wisconsin Pharmaceutical Association

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the Wisconsin Pharmaceutical Association text.

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Table of Contents Writing Chemical Formula Answers

1. Understanding the eBook Writing Chemical Formula Answers
 - The Rise of Digital Reading Writing Chemical Formula Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Writing Chemical Formula Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Writing Chemical Formula Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Writing Chemical Formula Answers
 - Personalized Recommendations
 - Writing Chemical Formula Answers User Reviews and Ratings
 - Writing Chemical Formula Answers and Bestseller Lists

5. Accessing Writing Chemical Formula Answers Free and Paid eBooks
 - Writing Chemical Formula Answers Public Domain eBooks
 - Writing Chemical Formula Answers eBook Subscription Services
 - Writing Chemical Formula Answers Budget-Friendly Options
6. Navigating Writing Chemical Formula Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Writing Chemical Formula Answers Compatibility with Devices
 - Writing Chemical Formula Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Writing Chemical Formula Answers
 - Highlighting and Note-Taking Writing Chemical Formula Answers
 - Interactive Elements Writing Chemical Formula Answers
8. Staying Engaged with Writing Chemical Formula Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Writing Chemical Formula Answers
9. Balancing eBooks and Physical Books Writing Chemical Formula Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Writing Chemical Formula Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Writing Chemical Formula Answers
 - Setting Reading Goals Writing Chemical Formula Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Writing Chemical Formula Answers
 - Fact-Checking eBook Content of Writing Chemical Formula Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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