

Word Equations

Write the skeleton equations for each of the following chemical reactions and then balance them:

1) When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made.



2) When isopropanol ($\text{C}_3\text{H}_8\text{O}$) burns in oxygen, carbon dioxide, water, and heat are produced.



3) When dissolved sodium hydroxide reacts with sulfuric acid, aqueous sodium sulfate, water, and heat are formed.



4) When fluorine gas is put into contact with calcium metal at high temperatures, calcium fluoride powder is created in an exothermic reaction.



5) When sodium metal reacts with iron (II) chloride, iron metal and sodium chloride are formed.



Writing Balanced Chemical Equations From Word Problems

William Levi Yarroch



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Balancing Chemical Equations Learning Hub Publishing, 2020-06-22 To improve their chemistry one must perfect the skill of balancing numerous chemical equations and doing them properly This is something which you will have to practice again and again to perfect as its a key skill when it comes to high school and even in college entrance exam and some college courses In this book you will find multiple practice equations with answers in 3 terms 4 terms 5 terms and many word problems that you can work on and improve you chemistry *Solving Problems in Chemistry* Richard G. Smith, Gary K. Himes, 1990 **General, Organic, and Biological Chemistry** Kenneth W. Raymond, 2013-01-04 General Organic and Biological Chemistry 4th Edition has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry An integrated approach is employed in which related general chemistry organic chemistry and biochemistry topics are presented in adjacent chapters This approach helps students see the strong connections that exist between these three branches of chemistry and allows instructors to discuss these interrelationships while the material is still fresh in students minds *Solving Problems in Chemistry* Smith, 1993-06

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