

Experiment 1

1. Write the general equation for an equilibrium/phase transition and give the conditions of each. Example: melting

Experiment 2: Liquid and Vapor Equilibrium in a Closed System

In a closed system, equilibrium is established between the liquid and the vapor of a substance. The conditions of equilibrium are:

- a. pressure is constant b. temperature is constant

a.	b.
a.	b.

Experiment 3: Vapor Pressure of a Pure Substance

1. Write the general equation for the vapor pressure of a pure substance. The conditions of equilibrium are:

- a. pressure is constant b. temperature is constant

c. volume is constant d. mass is constant

- e. pressure is constant f. temperature is constant

- g. volume is constant h. mass is constant

- i. pressure is constant j. temperature is constant

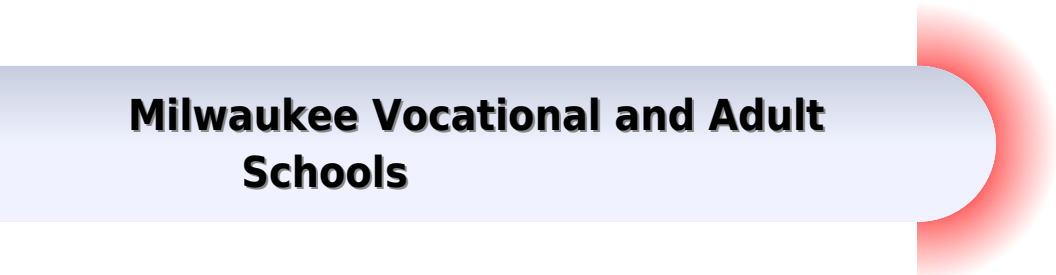
2. Write the general equation for the vapor pressure of a pure substance. The conditions of equilibrium are:

- a. pressure is constant

- b. temperature is constant

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