# **AP Chemistry Final Study Guide**

#### Unit 1: Atomic Structure and Properties (7%-9%)

- At STP, one mole of gas always occupies 22.4 L
- Mass Spectroscopy:
  - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
  - The most abundant isotope will have the greatest effect
- Coulomb's Law:
  - The greater the charge of the nucleus, the more energy is required to ionize an electron
  - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- · Electromagnetic Radiation:
  - When atoms absorb energy, electrons jump to higher energy levels
  - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- Kinetic Energy of Ionized Electrons:
  - The faster an ejected electron is going, the more kinetic energy it has
  - Electrons originally further away from the nucleus require less energy to eject, so
    it will have more kinetic energy
- · Electron Configuration:
  - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
    - Nickel: 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>6</sup>3d<sup>8</sup>4s<sup>2</sup>(notice the d-block)
    - Nickel (with shorthand): [Ar]4s<sup>2</sup>3d<sup>8</sup>
- Law of Definite Proportions:
  - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- Law of Multiple Proportions:
  - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- Aufbau Principle:
  - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

# **Spring Final Exam Study Guide Honors Chemistry**

University of Illinois at Urbana-Champaign. Student Senate

#### **Spring Final Exam Study Guide Honors Chemistry:**

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Kostecka, 2011-02-09 Morris William Travers A Lifetime of Achievement This book chronicles the remarkable life and accomplishments of Professor Morris William Travers It covers his entire life and showcases his accomplishments as a scientist educator administrator industrialist and author and encompasses the history of his family and his early life his work in setting up Bristol University India and setting up the Indian Institute of Science his work in England from 1914 to 1937 and the accomplishments and affairs of his later years In addition this book shows the visionary nature and ideas of Professor Travers and his impeccable sense of honor and integrity in dealing with others In describing this book I feel that it is best to do so by talking about different parts of the life of Professor Travers PART ONE The name Travers came from unknown origin though the first Travers in England may have been a chief in the Norman army and may have fought with William the Conqueror at the Battle of Hastings in 1066 There were also individuals with variations of the Travers name who settled in America and had some importance in politics of early America In England there were Travers family members who were bankers members of the Royal Society and even one who was a double agent during the war against Napoleon William Travers Morris s father was a well known doctor and one of the early practitioners of aseptic surgery while Morris s mother Anne came from a well to do family that had some naval and theological importance in England Morris born on January 24th 1872 and his siblings were cared for by a series of nannies nurses and maids Though William's medical practice was doing well his children lived simply Pocket money for the children was limited though William and Anne did take their family on many happy country holidays Morris received his early education at Ramsgate Woking College and at Blundell's School While at Blundell's he excelled in Chemistry but had significant difficulty with Latin He entered University College in London in 1891 It is likely that an interest in science for Morris Travers had a good deal to do with his father who always felt he had to read more about his field just to keep up and visits to the Regent Street Polytechnic and the Electrical Exhibition at the Crystal Palace He unlike most of his classmates enjoyed the weekly chemistry lectures he received at Blundell's William's encouragement was also important in Morris s choosing to study at University College in London Though he enjoyed zoology Morris soon fell under the tutelage of Ramsay and Plimpton He earned his B Sc in 1893 and began advanced study of chemistry later that same year Though he had entertained working with Haller in France to earn his Ph D Morris became fascinated with the discovery of a new gas by Ramsay helium and began work on the properties of the new gas He was invited by Ramsay to stay and look for the element with mass of 20 This effort led to the discovery of neon Further work with Ramsay and Hampson on liquefying the components of air brought about the discovery of first krypton and then xenon Morris who had great mechanical aptitude and some ability as a draftsman then was given the task of isolating hydrogen to separate neon from a mixture of air and helium He built an apparatus that so allowed the isolation of liquid and solid hydrogen to make such a separation for neon and then began a study of its properties He is first asked if he would be interested in being the Director of the proposed Indian institute of Science and publishes his first book The Experimental

Study of Gases He works on developing temperature scales by measuring the vapor pressures of liquid hydrogen and liquid oxygen meets his future wife Dorothy Gray and begins searching for an academi *Cincinnati Magazine*, 2009-11 Cincinnati Magazine taps into the DNA of the city exploring shopping dining living and culture and giving readers a ringside seat on the issues shaping the region *Lovejoy's College Guide*, 1993 **Universal Cyclopaedia and Atlas** Charles Kendall Adams, Rossiter Johnson, 1902 **Resources in Education**, 1990 **Peterson's Annual Guides to Graduate Study**, 1983 *The Official Guide to Catholic Educational Institutions and Religious Communities in the United States*, 1971

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