
AP CHEMISTRY FINAL EXAM REVIEW 2

1. Is energy absorbed or released in an endothermic reaction?: Absorbed.
2. What is the sign of delta H for an exothermic reaction?: Exothermic reactions release energy. The sign of delta H is negative.
3. When you touch a flask and it feels hot, is this an endothermic or exothermic process?: Exothermic; heat is being given off.
4. Define specific heat capacity.: The amount of heat required to raise 1g of a substance to 1 degrees Celsius.
5. $q = mc(\Delta T)$; What does "q" stand for?: Heat (kJ)
6. $q = mc(\Delta T)$; What does "c" stand for?: Specific heat capacity (J / g degrees C)
7. What is a hydrogen ion donor called?: An acid.
8. What is a hydrogen ion acceptor called?: A base.
9. Feels slippery; acid or base?: Base.
10. Tastes sour; acid or base?: Acid.
11. Tastes bitter; acid or base?: Base.
12. $\text{pH} < 7$; acid, base, or neutral?: Acid
13. $\text{pH} > 7$; acid, base, or neutral?: Base
14. Define neutralization.: When an acid reacts with a base to form water and a salt. $\text{pH} = 7$.
15. Soap; acid or base?: Base (tastes bitter)
16. Ammonia, acid or base?: Base (drain cleaners are bases).
17. Lemonade; acid or base?: Acid (tastes sour)
18. How many valence electrons are in an atom of chlorine?: 7
19. How many valence electrons does aluminum have?: 3
20. Define Reaction Rate: The decrease in concentration of reactants or the increase in concentration of products with time.
21. How does increasing temperature affect the rate of reaction?: Increasing temperature causes the kinetic energy to increase. Thus, molecules are moving faster and colliding more frequently. This causes the rate of reaction to increase.
22. If I decrease the concentration of reactant, what happens to the rate of reaction?: Decreasing concentration leads to less reactants available for collision. Therefore, reaction rate decreases.
23. How does surface area influence reaction rate?: By increasing the surface area of reactant molecules, we are increasing the space or point of contact that these molecules have available

Spring 2013 Chemistry Final Exam Answers

VM Jensen



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Paleo in a Nutshell Geoff Bond,2017-05-04 Nature has provided every creature that roams the earth a diet designed to sustain and maximize its life What these diets are composed of is based on a creature s location climate accessibility and physical prowess As humans our ancient forefathers and mothers were raised and nourished in the same manner as the beasts that surrounded them These Stone Age diets now called Paleo diets reflect the types of foods that nature intended us to eat Unfortunately we have lost touch with how nature intended us to live and mistakenly think we can eat anything sitting on our supermarket shelves Paleo in a Nutshell is an eye opening guide to restoring our health by eating the foods for which

our bodies are designed to consume Author Geoff Bond is a world renowned scientist who has extensively researched forager societies of the past and present What he has discovered is that our hunter gatherer ancestors had never experienced the health disorders that plague us today This is because their diets environments and daily activities were far more in touch with their bodies than our way of life today Although our diet has become our most dysfunctional feature other aspects of the Paleo lifestyle are essential to explore as well In Part One of this book we learn how and why our ancestors thrived and explore how things went wrong when we stopped behaving in agreement with our evolutionary past In Part Two we are provided with an Owner s Manual a guide to appropriate and healthy foods with which to fuel our bodies and a step by step plan for implementing this way of eating While you may no longer have time to hunt and forage for your food this book offers modern and simple substitutions for you to use as well as several delicious recipes Written in simple to understand language Paleo in a Nutshell is all you need to find your way back to a healthier and more natural way of living

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