

**Experiment 35****Advance Study Assignments: Spot Tests for Some Common Anions**

1. Each of the observations in the following list was made on a different solution. Given the observations, state which ion studied in this experiment is present. If the test is not definitive, indicate that with a question mark.
  - A. Addition of 6 M NaOH and Al to the solution produces a vapor that turns red litmus blue.  
Ion present: \_\_\_\_\_
  - B. Addition of 6 M HCl produces a vapor with an acid odor.  
Ion present: \_\_\_\_\_
  - C. Addition of 6 M HCl produces an effervescent.  
Ion present: \_\_\_\_\_
  - D. Addition of 6 M HNO<sub>3</sub>, plus 0.1 M AgNO<sub>3</sub>, produces a precipitate.  
Ion present: \_\_\_\_\_
  - E. Addition of 6 M HNO<sub>3</sub>, plus 1 M BaCl<sub>2</sub>, produces a precipitate.  
Ion present: \_\_\_\_\_
  - F. Addition of 6 M HNO<sub>3</sub>, plus 0.5 M (NH<sub>4</sub>)<sub>2</sub>MoO<sub>4</sub>, produces a precipitate.  
Ion present: \_\_\_\_\_
2. An unknown containing one or more of the ions studied in this experiment has the following properties:
  - A. No effect is observed on addition of 6 M HNO<sub>3</sub>.
  - B. No effect is observed on addition of 0.1 M AgNO<sub>3</sub> to the solution from Part A.
  - C. A white precipitate is observed on addition of 1 M BaCl<sub>2</sub> to the solution from Part A.
  - D. A yellow precipitate is observed on addition of (NH<sub>4</sub>)<sub>2</sub>MoO<sub>4</sub> to the solution from Part A.

On the basis of this information, which ions are present, which are absent, and which are in doubt?

Present

Absent

In doubt

# Spot Tests Advance Study Assignment

**Robert Ramotowski**



### **Spot Tests Advance Study Assignment:**

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