

Annotated periodic table of the elements

1	2											13	14	15	16	17	18		
1 H 1.01 Hydrogen												3 Li 6.94 Lithium	4 Be 9.01 Beryllium	5 B 10.81 Boron	6 C 12.01 Carbon	7 N 14.01 Nitrogen	8 O 16.00 Oxygen	9 F 18.99 Fluorine	10 Ne 20.18 Neon
3 Na 22.99 Sodium	4 Mg 24.31 Magnesium											11 Al 26.98 Aluminium	12 Si 28.09 Silicon	13 P 30.97 Phosphorus	14 S 32.06 Sulfur	15 Cl 35.45 Chlorine	16 Ar 39.95 Argon		
5 K 39.10 Potassium	6 Ca 40.08 Calcium											17 Ga 69.72 Gallium	18 Ge 72.64 Germanium	19 As 74.92 Arsenic	20 Se 79.60 Selenium	21 Br 79.90 Bromine	22 Kr 83.80 Krypton		
7 Rb 85.47 Rubidium	8 Sr 87.62 Strontium											19 In 76.90 Indium	20 Sn 118.71 Tin	21 Sb 121.76 Antimony	22 Te 127.60 Tellurium	23 I 126.91 Iodine	24 Xe 131.29 Xenon		
9 Cs 132.91 Cesium	10 Ba 137.33 Barium											21 Tl 204.38 Thallium	22 Pb 207.2 Lead	23 Bi 208.98 Bismuth	24 Po 209 Polonium	25 At 210 Astatine	26 Rn 222 Radon		
11 Fr 223 Francium	12 Ra 226 Radium											23 Ac 227 Actinium	24 Th 232.04 Thorium	25 Pa 231.04 Protactinium	26 U 238.03 Uranium	27 Np 237.05 Neptunium	28 Pu 244.06 Plutonium		
												25 Am 243.06 Americium	26 Cm 247.07 Curium	27 Bk 247.07 Berkelium	28 Cf 251.08 Californium	29 Es 252.08 Einsteinium	30 Fm 257.10 Fermium		
												27 Md 258.10 Mendelevium	28 Dn 261.10 Darmstadtium	29 Rg 261.10 Roentgenium	30 Nh 261.10 Nihonium	31 Fl 269.10 Flerovium	32 Lv 260.10 Livermorium		
												29 Ts 261.10 Tennessine	30 Og 261.10 Oganesson						

Most common oxidation states, electronegativity (Mulliken-Pauling)

Atomic number and symbol

Atomic weight and name

1
H
1.01
Hydrogen

3
Li
6.94
Lithium

5
Na
22.99
Sodium

7
K
39.10
Potassium

9
Cs
132.91
Cesium

11
Fr
223
Francium

2
He
4.003
Helium

4
Be
9.012
Beryllium

6
Mg
24.31
Magnesium

8
Ca
40.08
Calcium

10
Sr
87.62
Strontium

12
Ba
137.33
Barium

14
Ra
226
Radium

15
Scandium

16
Titanium

17
Vanadium

18
Chromium

19
Manganese

20
Iron

21
Cobalt

22
Nickel

23
Copper

24
Zinc

25
Gallium

26
Germanium

27
Arsenic

28
Selenium

29
Bromine

30
Krypton

31
Rubidium

32
Strontium

33
Yttrium

34
Zirconium

35
Niobium

36
Molybdenum

37
Technetium

38
Ruthenium

39
Rhodium

40
Palladium

41
Silver

42
Cadmium

43
Indium

44
Tin

45
Antimony

46
Tellurium

47
Iodine

48
Xenon

49
Barium

50
Lanthanum

51
Cerium

52
Praseodymium

53
Neodymium

54
Promethium

55
Samarium

56
Europium

57
Gadolinium

58
Terbium

59
Dysprosium

60
Holmium

61
Erbium

62
Thulium

63
Ytterbium

64
Lutetium

65
Hafnium

66
Tantalum

67
Tungsten

68
Rhenium

69
Osmium

70
Iridium

71
Platinum

72
Gold

73
Mercury

74
Thallium

75
Lead

76
Bismuth

77
Polonium

78
Astatine

79
Radon

80
Francium

81
Radium

82
Actinium

83
Thorium

84
Protactinium

85
Uranium

86
Neptunium

87
Plutonium

88
Americium

89
Curium

90
Berkelium

91
Californium

92
Einsteinium

93
Fermium

94
Mendelevium

95
Nobelium

96
Lawrencium

97
Rutherfordium

98
Dubnium

99
Seaborgium

100
Bohrium

101
Hassium

102
Meitnerium

103
Darmstadtium

104
Roentgenium

105
Copernicium

106
Nihonium

107
Flerovium

108
Livermorium

109
Tennessine

110
Oganesson

Group name*

*Groups 1 to 10 are named after their first members i.e. Group 1 is the Scandium group

1 Hydrogen & the alkali metals

2 Alkaline earth metals

3 Coinage metals

4 Volatile metals

5 Lanthanides

6 Crystallogens

7 Pnictogens

8 Chalcogens

9 Halogens

10 Helium group

- Group names*
- Hydrogen & the alkali metals
 - Alkaline earth metals
 - Coinage metals
 - Volatile metals
 - Isotopes
 - Chalcogens
 - Chalcogens
 - Halogens
 - Helium group

*Groups 3 to 10 are named after their first members i.e. Group 3 is the Scandium group

Main group
oxidation states
electronegativity
(Alred-Pauling)

Atomic number
and symbol
Atomic weight
and name

CLASS COUNTERPARTS

Metals	Nonmetals
Pre-transition	
Active	Active
Transition	Intermediate
Post-transition	Metals
Metals (Ru-Pt, Os-Au)	Metals

Active metals

- Pre-transition metals**
Light, reactive, mostly of low mechanical strength, melting and boiling points. The group 1 members are easily cut with a knife. Only Be, Mg and Al have any structural uses. Most form basic oxides (Be and Al are amphoteric).
- Rare earths**
Closely resemble Ca, Sr and Ba, but many are heavier and they form mostly pale coloured compounds. Similar to one another, esp. La to Lu (aka the lanthanides), and hard to separate for that reason. Abundant but widely dispersed; commercially viable concentrations are rare. Oxides are strongly to moderately basic. Sc, Yb and Lu exhibit weak amphotericism.
- Actinides**
Soft, dense and reactive metals. Those from Th onwards are synthetic. Early actinides (Th-Am) show some similarities to transition metals and have basic, amphoteric, or acidic oxides; late actinides are more like lanthanides. Many actinides form coloured compounds.

Transition metals

- Moderate to high density metals, with high melting and boiling points, many have high hardness and mechanical strength. Those from Ir onwards are synthetic. Chemically show variable valency and strong tendency to form coordination and colored compounds. Oxides are basic, amphoteric or acidic, depending on oxidation state.
- Noble metals**
Resistant to corrosion and oxidation, unlike base metals such as Fe, Ni, Cu and Zn. Commonly found together in native form. Good catalytic properties. Very easily electroplated; insoluble oxides.
- Frontier metals**
Soft (or brittle), low strength, and with melting points lower than transition metals. Crystalline structures tend to show directional bonding, with generally greater complexity or lower nearest neighbours than other metals. All used or tried as soft metal lubricants. Most are chemically weak; a minority are moderately electropositive. Characterised, to varying degrees, by covalent bonding tendencies, amphotericism, and anionic species such as zincates, stannates, and bismuthates.

Metals

- Metalloids**
Metallic looking brittle, fair electrical conductors. All known in semiconducting forms. Moderate ionisation energies (750–1,000 kJ/mol), and electronegativity values (1.8–2.3). Mostly behave chemically as non-metals. Can form alloys with metals. Amphoteric or weakly acidic oxides.
- Intermediate nonmetals**
H and N are colorless gases; C, P, and Se are metalloids looking solids while S is yellow. The latter four are brittle but all known in malleable, plastic or ductile forms. Moderate to high ionisation energies, low to high electron affinities, moderate to high electronegativity values, relatively poor to moderately strong oxidising agents, tend to form covalent or metallic to near-metallic compounds with metals. Their oxides are acidic, or neutral (H₂O, CO, NO, N₂O).
- Group 1**
The alkali metals Li, Na, K, Rb, Cs, and Fr are soft, silvery-white, and highly reactive. They are the most reactive elements in the periodic table. They are found in nature only as compounds, and are highly reactive with water and air. They are the most reactive elements in the periodic table.

Nonmetals

- Chalcogens**
O, S, Se, and Te are, respectively, colorless, pale yellow, and pale yellow-green gases. Br is a red-brown liquid. I is a metallic looking solid. Chemically tend to have high ionisation energies, electron affinities, and electronegativity values, and be relatively strong oxidising agents. Manifestations include oxygen's major association with the ubiquitous processes of corrosion and combustion, and the intrinsically corrosive nature of the non-metallic halogens. All five tend to form predominantly ionic compounds with metals. Their oxides are acidic.
- Noble gases**
Colorless, odorless, non-flammable gases with very low chemical reactivity. The first noble gas compound, of approximate composition XePF₆, was not prepared until 1962. More than half-a-thousand noble gas compounds were synthesised in the ensuing half-century. Compounds of He, Ne, and Ar, in ambient conditions, are not known.

Radioactive
Oganesson is likely to be either a metalloid or a post-transition metal

Who Am I Answer Key Periodic Table

Jan Rader, Jan Ray



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Comprehensive Chemistry XI Dr. B. Kapila, S. K. Khanna, 2010-11 Comprehensive chemistry according to the new syllabus prescribed by Central Board of Secondary Education CBSE *Science and Technology Handbook Part 2* Chandan Sengupta, A Workbook for CBSE Students of Class X Chandan Sengupta Title Science and Technology Handbook Part 2 Author Chandan Sengupta Year of Publication 2025 Date of Publication 14 02 2025 Format Book Country INDIA ISBN 978 93 342 2360 6 This book has been published with all reasonable efforts duly taken up to make the material error free after getting the consent of the author No part of this book shall be used reproduced in any manner whatsoever without written permission from the author except in the case of brief quotations embodied in critical articles and reviews What we expect from our fellow student and what are the facilities we provide them should have proper links for ensuring the maximum return of our efforts We even come across instances during which children may revolt repeatedly to avoid scheduled intensive learning programmes duly planned for them For efficient handling of such jobs we should go on planning content delivery mechanism on the basis of student centered curriculum design It will even link up our plan with those of other fellow faculty members for making the effort a vibrant one The work book like this one and others of similar category has a comprehensive plan of addressing content areas duly specified by the boards of studies Answer sheets are there for some selected test papers Rest of the other sheets kept off the side for enabling the exploratory drive of fellow students active We are expecting their active participation in the learning and facilitation drives It is true that this workbook cannot follow the content areas exclusively prescribed for the aspirants of the particular age group The purpose of the incorporations of varying types of activities is to expose the fellow students to some forthcoming challenges It will definitely imply a sort of impression in the mind of the student and enable them to grasp through higher challenges with subtle easiness

Microsoft Word Simple Projects Jan Rader, Jan Ray, 2001-04 Projects for language arts social studies science and math Provided templates can be modified to meet specific needs Project samples also provided [Life in the Universe, 5th Edition](#) Jeffrey Bennett, Seth Shostak, Nicholas Schneider, Meredith MacGregor, 2022-08-23 The world's leading textbook on astrobiology ideal for an introductory one semester course and now fully revised and updated Are we alone in the cosmos How are scientists seeking signs of life beyond our home planet Could we colonize other planets moons or even other star systems This introductory textbook written by a team of four renowned science communicators educators and researchers tells the amazing story of how modern science is seeking the answers to these and other fascinating questions They are the questions that are at the heart of the highly interdisciplinary field of astrobiology the study of life in the universe Written in an accessible conversational style for anyone intrigued by the possibilities of life in the solar system and beyond Life in the Universe is an ideal place to start learning about the latest discoveries and unsolved mysteries in the field From the most recent missions to Saturn's moons and our neighboring planet Mars to revolutionary discoveries of thousands of exoplanets

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Surfactants in Solution K.L. Mittal, 2012-12-06 This and its companion volumes 8 9 and 10 document the proceedings of the 6th International Symposium on Surfactants in Solution SIS held in New Delhi India August 18 22 1986 under the joint auspices of the Indian Society for Surface Science and Technology and Indian Institute of Technology Delhi As this symposium was a landmark it represented the tenth anniversary of this series of symposia so it is very apropos to reflect on how these symposia have evolved to their present size and status The pedigree of this series of symposia goes back to 1976 when the premier symposium in this series was held Actually in 1976 it was a modest start and it was not possible at that time to gaze at the crystal ball and predict what would be the state of affairs in 1986 For historical purposes it should be recorded here that the first symposium was held in Albany NY under the title Micellization Solubilization and Microemulsions the second symposium was christened Solution Chemistry of Surfactants and was held in Knoxville TN in 1978 the venue for the third symposium in 1980 was Potsdam NY and it was dubbed International Symposium on Solution Behavior of Surfactants Theoretical and Applied Aspects

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Visual Literacy Mark Newman, Donna Ogle, 2019-10-31 Visual Literacy examines how teachers can use visuals to improve learning for all students It provides teachers with a foundation in visual literacy defined as the ability to read think and communicate with visually presented information Results of studies of students using visual information indicate that most students are clearly lacking in the tools needed to use visuals effectively The book orients teachers to visual literacy and the world of visuals It discusses various classroom tested strategies and activities for all students including second language learners and students with special needs Stressing visual literacy skills helps students understand a visual more deeply so they can master the content they are learning Teachers will learn to employ a literacy triad of reading thinking and communicating to aid students in their study of visuals First they inquire into the visual reading it for content and context including assessing the authenticity of the document Second they

think about the document by analyzing interpreting and evaluating it to come up with answers to their inquiry Graphic organizers help students decipher the content and understand the meaning of the visual document connecting it to prior and future instruction Third they communicate their findings using visuals *Wild Things They Don't Tell Us - Aliens, Alchemy, Government Denials - The Truth is in Here!* Reg Presley, 2014-02-28 Are you one of those people who when someone tells you they've seen a UFO think that they must be one can short of a six pack Do you still believe there has been no proof in all these years as to their existence If so then think again One summer's day Reg Presley whose band The Troggs shot to fame in the sixties with unforgettable songs such as Wild Thing and Love Is All Around Us walked into a crop formation that changed his life forever Through remarkable meetings with crop circle and UFO investigators eyewitnesses military personnel scientists astronomers and astronauts among others Reg was to discover a world behind anything he had ever experienced a world of strange patterns in corn fields inexplicable craft shooting through our skies alien abduction historical anomalies suppressed information establishment denial and disinformation Never one to resist a puzzle Reg has spent the past twelve years researching this phenomena No stone has been left unturned no dark corner left unexplored Rock star and visionary reg is the man who really knows the truth The Sunday Times The Whole-brain Solution Tricia Armstrong, 2003 Explores the higher order thinking tools that are essential for students to become effective learners It includes lessons that encourage students to understand and integrate information so that they can use what they know to solve problems and make decisions

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traditional courses and the recent innovations in the field of basic Chemistry have been incorporated. The books contain a large number of worked out examples, illustrations, illustrative questions, numerical problems, figures, tables, and graphs.

Chemistry Greg Curran, 2011. Covers all the topics in a typical one year high school chemistry curriculum. **The IT in Secondary Science Book** Roger Frost, 1994. MCAT General Chemistry Review 2023-2024 Kaplan Test Prep, 2022-07-05. Kaplan's MCAT General Chemistry Review 2023-2024 offers an expert study plan, detailed subject review, and hundreds of online and in-book practice questions, all authored by the experts behind the MCAT prep course that has helped more people get into medical school than all other major courses combined. Prepping for the MCAT is a true challenge; Kaplan can be your partner along the way, offering guidance on where to focus your efforts and how to organize your review. This book has been updated to match the AAMC's guidelines precisely, so no more worrying about whether your MCAT review is comprehensive. The Most Practice: More than 350 questions in the book and access to even more online, more practice than any other MCAT general chemistry book on the market. The Best Practice: Comprehensive general chemistry subject review is written by top-rated, award-winning Kaplan instructors. Full color 3-D illustrations from Scientific American, charts, graphs, and diagrams help turn even the most complex science into easy-to-visualize concepts. All material is vetted by editors with advanced science degrees and by a medical doctor. Online resources, including a full-length practice test, help you practice in the same computer-based format you'll see on Test Day. Expert Guidance: High-yield badges throughout the book identify the topics most frequently tested by the AAMC. We know the test: The Kaplan MCAT team has spent years studying every MCAT-related document available. Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test.

Secondary Starters and Plenaries Kate Brown, 2013-04-14. Starters and plenaries are now established elements of all good lesson planning. A good starter gets a class engaged right from the word go, challenges and motivates students, and sets a positive tone for the rest of the lesson. A good plenary allows students to focus on the key objectives of the lesson and to reflect on the progress they have made. Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02. Chemistry: The Molecular Nature of Matter, 8th Edition, continues to focus on the intimate relationship that exists between structure at the atomic/molecular level and the observable macroscopic properties of matter. Key revisions in this edition focus on three areas: The deliberate inclusion of more updated real-world examples that relate common real-world student experiences to the science of chemistry. Simultaneously, examples and questions have been updated to align them with career concepts relevant to the environmental, engineering, biological, pharmaceutical, and medical sciences. Providing students with transferable skills with a focus on integrating metacognition and three-dimensional learning into the text. When students know what they know, they are better able to learn and incorporate the material. Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment, answer-specific responses, and additional practice resources. The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high-level learning and increase

retention of chemistry knowledge Problems are arranged in an intuitive confidence building order Science John Michels (Journalist),1924 *Proceedings of the Second International Conference on Computer and Communication Technologies* Suresh Chandra Satapathy,K. Srujan Raju,Jyotsna Kumar Mandal,Vikrant Bhateja,2015-09-03 The book is about all aspects of computing communication general sciences and educational research covered at the Second International Conference on Computer Communication Technologies held during 24-26 July 2015 at Hyderabad It hosted by CMR Technical Campus in association with Division V Education Research CSI India After a rigorous review only quality papers are selected and included in this book The entire book is divided into three volumes Three volumes cover a variety of topics which include medical imaging networks data mining intelligent computing software design image processing mobile computing digital signals and speech processing video surveillance and processing web mining wireless sensor networks circuit analysis fuzzy systems antenna and communication systems biomedical signal processing and applications cloud computing embedded systems applications and cyber security and digital forensic The readers of these volumes will be highly benefited from the technical contents of the topics

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