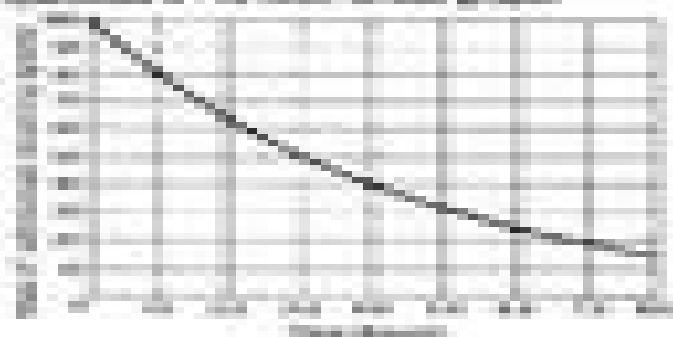


2-4- Nuclear Chemistry

APRACTICE TEST

1. Silver-107 (natural) was first known to be:
 - a) a transmuting element
 - b) causing the first "radioactivity"
 - c) a transmuting radioactivity
 - d) causing radioactivity
2. Silver-107 (natural) was first known to be:
 - a) a transmuting element
 - b) causing silver radioactivity
 - c) a transmuting element
 - d) a constant natural source
3. Radio-108 (radioactive "Silver-108") will be:
 - a) a beta emitter
 - b) an alpha emitter
 - c) a gamma emitter
 - d) a positron emitter
4. Which of the change caused by alpha:
 - a) $A \rightarrow A-4$ $Z \rightarrow Z-2$ $A \rightarrow A-4$ $Z \rightarrow Z-2$
 - b) $A \rightarrow A-4$ $Z \rightarrow Z-2$ $A \rightarrow A-4$ $Z \rightarrow Z-2$
 - c) $A \rightarrow A-4$ $Z \rightarrow Z-2$ $A \rightarrow A-4$ $Z \rightarrow Z-2$
 - d) $A \rightarrow A-4$ $Z \rightarrow Z-2$ $A \rightarrow A-4$ $Z \rightarrow Z-2$
5. Which of the following is correct in alpha:
 - a) alpha
 - b) gamma
 - c) beta
 - d) positron
6. What will the new composition of a substance with a half-life of 1.5 hours, which was at the following composition in the first 1.5 hours:
 - a) 100% of the substance
 - b) 75% of the substance
 - c) 50% of the substance
 - d) 25% of the substance
7. In a decay series of the (226) the first three steps involve alpha decay and then beta decay. What is the result of these steps?
 - a) 226-88
 - b) 226-88
 - c) 226-88
 - d) 226-88

Question 8 - Refer to the graph:



8. According to the above data, what is the half-life of the substance?
 - a) 10 hours
 - b) 25 hours
 - c) 50 hours
 - d) 100 hours
9. What percent of the original sample remains after 5 hours?
 - a) 100%
 - b) 75%
 - c) 50%
 - d) 25%
10. What is the half-life of the substance?
 - a) 10 hours
 - b) 25 hours
 - c) 50 hours
 - d) 100 hours
11. Radio-108 has a half-life of 1.5 hours. What percent of a sample remains after 1.5 hours?
 - a) 100%
 - b) 75%
 - c) 50%
 - d) 25%
12. Which of the following describes what occurs in the decay process?
 - a) is a decay series in which the substance is transformed into a stable element
 - b) is a decay series in which the substance is transformed into a stable element
 - c) is a decay series in which the substance is transformed into a stable element
 - d) is a decay series in which the substance is transformed into a stable element

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