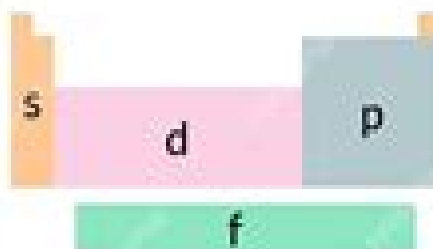


Preiodic Table of the Elements

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
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87 Fr	88 Ra	89 Ac	90 Rf	91 Db	92 Sg	93 Bh	94 Hs	95 Mt	96 Ds	97 Rg	98 Cn	99 Nh	100 Fl	101 Mc	102 Lv	103 Ts	104 Og
Lanthanoid series		73 Ce	74 Pr	75 Nd	76 Pm	77 Sm	78 Eu	79 Gd	80 Tb	81 Dy	82 Ho	83 Er	84 Tm	85 Yb	86 Lu		
Actinoid series		90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr		

s p d f Blocks

- Increasing left to right
- ↓ Increasing up to down
- ↗ Decreasing left to right
- ↑ Decreasing up to down



electronegativity	→ ↑
electropositivity	→ ↓
Atomic Radius / size	→ ↓
Metallic Properties	→ ↓
Non-metallic Properties	→ ↑
Ionization Energy	→ ↑
Electron Affinities	→ ↑

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Didier Musso



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Chapter 18 explains what the element of Fire is Chapter 19 explains what the element of Air is Chapter 20 explains what the element of Water is Chapter 21 explains what the element of Earth is Chapter 22 explains the true meaning of Saint Bartholomew Chapter 23 explains all about hexagonal growth formation in nature as mathematical construct Antimony, Gold, and Jupiter's Wolf Peter Wothers,2019 How did the elements get their names The origins of californium may be obvious but what about oxygen Investigating their origins takes Peter Wothers deep into history Drawing on a wide variety of original sources he brings to light the astonishing the unusual and the downright weird origins behind the element names we take for granted Scientifica Essentials ,2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

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