

Solubility Rules

Soluble compounds include ionic compounds containing ...

1. ... alkali metal ions (such as Na^+ , K^+ , etc.).
2. ... the ammonium ion (NH_4^+).
3. ... chloride, bromide, or iodide ions **EXCEPT** when paired with Ag^+ , Pb^{2+} , or Hg_2^{2+} ions.
4. ... nitrate, acetate, or chlorate ions **EXCEPT** $\text{AgC}_2\text{H}_3\text{O}_2$.
5. ... the sulfate ion **EXCEPT** when paired with Pb^{2+} , Ca^{2+} , Sr^{2+} , Ba^{2+} , or Ra^{2+} ions.

Insoluble compounds include ionic compounds containing ...

6. ... the sulfide ion **EXCEPT** when paired with alkali metal, alkaline earth metal, or ammonium ions.
7. ... the hydroxide ion **EXCEPT** when paired with alkali metal, ammonium, or Sr^{2+} , or Ba^{2+} ions.
8. ... phosphate, carbonate, or sulfite ions **EXCEPT** when paired with alkali metal or ammonium ions.

Solubility Rules Pogil Answers

U. Benedict



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Solubility Rules Based on Lattice Parameter Differences U. Benedict, 1977 **The Experimental Determination of Solubilities** G. T. Hefter, R. P. T. Tomkins, 2003-08-01 In diesem Band werden vielfältige experimentelle Verfahren zur Bestimmung von Löslichkeiten beschrieben. Commission on Solubility Data der IUPAC beschloß die Qualität der Löslichkeitsdaten international anzugleichen und zu verbessern indem verlässliche Verfahren zur Ermittlung dieser Daten zusammengefaßt veröffentlicht werden. Alle Datentypen wurden hier berücksichtigt ausführliche Angaben zur Zuverlässigkeit der einzelnen Verfahren und zur Auswahl der jeweils geeigneten Methode.

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