The image below represents the solubility curves of several compounds. Answer the given questions.



- 1. List the substances whose solubility decreases as temperature increases.
- 2. Which substance is least affected by temperature changes?
- 3. How many grams of ammonium chloride (NH₄Cl) forms a saturated solution at 50°C? _____
- 4. Which two substances have the same solubility at approximately 78°C?
- 5. Which compound is least soluble in water at 10°C?
- 6. What is the solubility of KCl at 25 °C? _____
- 7. What is the solubility of Ce2(SO4), at 50 °C? ______
- 8. How many grams of KNO3 can be dissolved at 50°C? _____

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James Kendall

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book contains numerous solved problems and exercises which bridge the gap between theoretical learning and practical implementation All the numerical problems are solved with block diagrams to reinforce the understanding of the concepts Primarily intended as a text for the undergraduate students of chemical engineering it will also be useful for other allied branches of chemical engineering such as polymer science and engineering and petroleum engineering KEY FEATURES Methods of calculation for stoichiometric proportions with practical examples from the Industry Simplified method of solving numerical problems under material balance with and without chemical reactions Conversions of chemical engineering equations from one unit to another Solution of fuel and combustion and energy balance problems using tabular column

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