

Model 2 – Ternary Acids (Oxyacids)

Acid	Name of Acid in Aqueous Solution	Cation (+)	Polyatomic Anion (-)	Polyatomic Anion Name
HClO_3	Chloric acid	H_3O^+	ClO_3^{-1}	Chlorate
H_2SO_3	Sulfurous acid	$2\text{H}_3\text{O}^+$	SO_3^{-2}	Sulfite
H_2SO_4	Sulfuric acid	$2\text{H}_3\text{O}^+$	SO_4^{-2}	Sulfate
H_3PO_3	Phosphorous acid	$3\text{H}_3\text{O}^+$	PO_3^{-3}	Phosphite
H_3PO_4	Phosphoric acid	$3\text{H}_3\text{O}^+$	PO_4^{-3}	Phosphate
HNO_3	Nitric acid	H_3O^+	NO_3^{-1}	Nitrate
HNO_2	Nitrous acid	H_3O^+	NO_2^{-1}	Nitrite
H_2CO_3	Carbonic acid	$2\text{H}_3\text{O}^+$	CO_3^{-2}	Carbonate

6. Look at the formulas of the ternary acids in Model 2.

a. How are ternary acids different from binary acids in their structure?

Ternary acids contain three elements and they also contain the element oxygen.

b. What number do you think the prefix "ter-" refers to?

"ter" refers to three

7. When ternary acids are mixed with water, ions will form. Fill in the table above with the formulas and names of the anions.

8. Examine the pairs of ternary acids in Model 2 that contain sulfur, phosphorus, and nitrogen. Each pair has one acid that ends in "-ic" and another that ends in "-ous." These endings are related to the name of the polyatomic anion found in the acid ("-ate" or "-ite"). Complete the statements below with the correct acid name ending.

Polyatomic anion ending is "-ate" → acid name ending is "ic"

Polyatomic anion ending is "-ite" → acid name ending is "ous"

9. If the prefix "hydro-" were used to name a ternary acid, what problem would this create when naming HClO_3 ?

ClO_3^{-1} has the name of Chlorate. "ate" must be replaced with "ic". If "hydro" is used as a prefix, it would be hydrochloric acid. This has the formula HCl , not HClO_3 .

10. Write a rule for naming ternary acids.

- The prefix "hydro" is not used in the name when an acid contains more than two elements and one of the elements is oxygen.

- "ate" is replaced with "ic" and "ite" is replaced with "ous".

11. Predict the formula for chlorous acid.



12. Circle the acid(s) below that would be named beginning with the prefix "hydro-."



Solubility Answer Key Pogil

Joel Henry Hildebrand



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Solutions and Solubilities Michael R. J. Dack, 1975 **Principles of Solution and Solubility** Kozo Shinoda, 1991
Solubilities of Inorganic and Organic Substances, 1907 Solubilities of Inorganic and Organic Substances, 1907
Solubilities of Inorganic and Organic Compounds Atherton Seidell, 1919 **The Experimental Determination of Solubilities** G. T. Hefter, R. P. T. Tomkins, 2003-08-01 In diesem Band werden vielfältige experimentelle Verfahren zur Bestimmung von Löslichkeiten beschrieben. Commission on Solubility Data der IUPAC beschloß die Qualität der Löslichkeitsdaten international anzugleichen und zu verbessern indem verlässliche Verfahren zur Ermittlung dieser Daten zusammengefaßt veröffentlicht werden. Alle Datentypen wurden hier berücksichtigt. Ausführliche Angaben zur Zuverlässigkeit der einzelnen Verfahren und zur Auswahl der jeweils geeigneten Methode.
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Walters,1991-12-10 This book is a compilation of experimentally determined solubility ranges of over 1 700 compounds in the National Toxicology Program s Chemical Repository Each compound s solubility was determined in a consistent manner in one to six solvents Solvents chosen were those most commonly used for toxicology studies spill cleanups and chemical synthesis or chemical reaction experiments These solvents include acetone 95% ethanol water dimethyl sulfoxide methanol and toluene Data for many of the research and industrial chemicals featured in this volume do not exist anywhere else If you are a toxicologist safety professional industrial hygienist or chemist this book is a valuable reference tool you ll find yourself using every day *Solubility curves* Charles Hertzel Frantz,1933 Solubility Joel Henry Hildebrand,1924

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