

SQA N5 Scotland Chemistry paper 2 2025

(i) Calculate the average rate of reaction, in $\text{cm}^3 \text{s}^{-1}$, for the first 50 seconds. 2

Show your working.

$$R = \frac{\Delta \text{time}}{\Delta \text{amount}}$$

Ans. $1.7 \text{ cm}^3 \text{s}^{-1}$

$\Delta = 0.85 \text{ cm}^3$

(ii) Add a curve to the graph to show the effect of the experiment. 1

(An additional graph is provided.)

line

(iii) A similar experiment was carried out using zinc powder and the same volume of sulfuric acid. 1

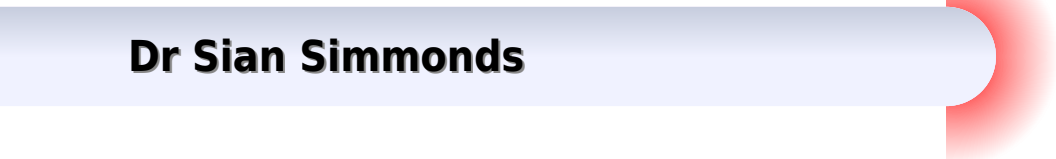
Predict the volume of gas produced.



* X 8 1 3 7 5 0 1 0 7 *

Sqa 2higher Chemistry Paper

Dr Sian Simmonds



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