AP Chemistry Final Study Guide

Unit 1: Atomic Structure and Properties (7%-9%)

- At STP, one mole of gas always occupies 22.4 L
- Mass Spectroscopy:
 - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
 - The most abundant isotope will have the greatest effect
- Coulomb's Law:
 - The greater the charge of the nucleus, the more energy is required to ionize an electron
 - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- · Electromagnetic Radiation:
 - When atoms absorb energy, electrons jump to higher energy levels
 - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- Kinetic Energy of Ionized Electrons:
 - The faster an ejected electron is going, the more kinetic energy it has
 - Electrons originally further away from the nucleus require less energy to eject, so
 it will have more kinetic energy
- · Electron Configuration:
 - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
 - Nickel: 1s²2s²2p⁶3s²3p⁶3d⁸4s²(notice the d-block)
 - Nickel (with shorthand): [Ar]4s²3d⁸
- Law of Definite Proportions:
 - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- Law of Multiple Proportions:
 - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- Aufbau Principle:
 - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

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Lei Shi

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Environmental Organic Chemistry for Engineers James G. Speight, 2016-11-02 Environmental Organic Chemistry for Engineers clearly defines the principles of environmental organic chemistry and the role they play in forming remediation strategies In this reference the author explores parameter estimation methods the thermodynamics and kinetics needed to predict the fate transports and reactivity of organic compounds in air water and soils The book s four part treatment starts with the classification of organic molecules and physical properties of natural organic matter halocarbons phenols polyaromatic hydrocarbons organophosphates and surfactants An overview of remediation technologies and a discussion of the interactions that lead to physical properties that affect chemical distribution in the environment is also detailed as are the important reaction classes of organic molecules including substituent effects and structure and activity relationships found in Part Two and Three Part four is devoted to the strengths and weaknesses of different remediation technologies and when they should be employed Clearly defines the principles of environmental organic chemistry and the role they play in forming remediation strategies Includes the tools and methods for classifying environmental contaminants found in air water and soil Presents a wide range of remediation technologies and when they should be deployed for maximum effect

Bulletin of Information United States Coast Guard Academy,
Wilson-Kennedy, Goldie S. Byrd, Eugene Kennedy, Henry T. Frierson, 2019-02-28 This book reports on high impact educational practices and programs that have been demonstrated to be effective at broadening the participation of underrepresented groups in the STEM disciplines

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<u>Undergraduate Catalog ... with Announcements for ...</u> Western Michigan University,1929 **University Record**,1903 **How Finns Learn Mathematics and Science**,2007-01-01 The Finnish students success in the first PISA 2000

evaluation was a surprise to most of the Finns and even people working in teacher education and educational administration had difficulties to believe that this situation would continue Finland's second success in the next PISA 2003 comparison has been very pleasing for teachers and teacher educators and for education policymakers. The good results on the second time waked us to think seriously on possible reasons for the success Several international journalists and expert delegations from different countries have asked these reasons while visiting in Finland Since we had no commonly acceptable explanation to students success we decided at the University of Helsinki to put together a book How Finns Learn Mathematics and Science in order to give a commonly acceptable explanation to our students success in the international PISA evaluations The book tries to explain the Finnish teacher education and school system as well as Finnish children's learning environment at the level of the comprehensive school and thus give explanations for the Finnish PISA success The book is a joint enterprise of Finnish teacher educators The explanations for success given by altogether 40 authors can be classified into three groups Teacher and teacher education school and curriculum and other factors like the use of ICT and a developmental project LUMA The main result is that there is not one clear explanation although research based teacher education seems to have some influence But the true explanation may be a combination of several factors Basics of Ecotoxicology Donald W. Sparling, 2017-07-28 This textbook presents a comprehensive examination of environmental science and ecotoxicology for undergraduate students The material provides sufficient related background information leading to a competency to clearly understand ecotoxicology concepts and topics Selected Water Resources Abstracts, 1989 **Host Bibliographic** Record for Boundwith Item Barcode 30112113333022 and Others ,1921 Inventory of Federal Energy-related Environment and Safety Research for ... ,1980 Inventory of Federal Energy-related Environment and Safety Research for FY 1979, 1980 Catalogue Number. Course Catalog Anonymous, 2025-08-07 Reprint of the original first published in 1876 The Antigonos publishing house specialises in the publication of reprints of historical books We make sure that these works are made available to the public in good condition in order to preserve their cultural heritage Spring Term Announcements Indiana University, 1909

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