

Sports and Entertainment Marketing

School Name _____
 Teacher Name _____
 Teacher Contact Information _____

This Marketing course is designed to develop skills in the areas of marketing analysis, event/marketing, communications and human relations along with a thorough understanding of the sports and entertainment industry and career options available. Course objectives include but are not limited to:

Marketing Program Overview
 Developing Job Search Skills
 Understanding the Importance of SEM Marketing
 Developing an Overview of Marketing
 Developing Effective Communication Strategies
 Careers in the Sports & Entertainment Field
 Describe the Economics of the Sports Industry
 Describing Product Development
 Workplace Readiness

Exploring the Concept of Branding
 Describing the Product Development
 Exploring Pricing & Distribution Strategies
 Developing a Promotional Mix
 Examining Product Licensing
 Exploring Forms of Advertising
 Using Social Media
 Exploring Public Relations & Sales Promotion
 Exploring Sponsorship & Endorsements
 Exploring the Marketing Plan

Course Evaluation: One letter grade will be posted on the Google Classroom and will be communicated verbally.

Classroom Management: Students will be given assignments in class daily.

Quizzes/Tests: Quizzes/Tests will consist of a variety of question types including multiple choice, true/false, matching, filling, and essay/question.

Project: There will be a major project, both independent and group, during the semester that relate to the course objectives. Ample time will be given in class for work to be completed. It is important to use this time wisely.

OECFA: OECFA is our on curriculum organization for marketing students and complementary classroom instruction. On occasion, students will be expected to participate in OECFA activities during class time for course credit. *Students may opt-in to join OECFA membership for playing local, state, and national events. This year, OECFA dues will be \$100.00. Students must be paid members in order to participate in outside-of-class OECFA activities including but not limited to certain field trips, competitions, and "club day" activities. Check playmate information. *OECFA participation is a privilege and students will be in good academic standing all semester to be eligible for activities outside of school. Specifically, students will have a 1.00 average in all classes while in a club trip and must have attendance taken in order to be eligible for OECFA activities.

Final Assessment: Students will spend a large portion of the semester working on a project to evaluate student learning, skill attainment, and achievement. This project will include meaningful, authentic assignments that reflect the real-world experience of business and marketing professionally. At final evaluation of this project will take place near the end of the semester to assess the students' cumulative learning in the course. *Students are also expected to be professional in their behavior and conduct for the project evaluation. Final assessment will be only used if the teacher the student's final grade for the course.

Google Classroom and Drive: We will be using the Google platform for accessing class material and turning in assignments. The Google Classroom code is _____.

Calculation of Missing Grade:

I use the highest possible standard 100 grading scale and my gradebook will operate on a "Total Points System." Grades will be updated as Parents have a concern.

Test: 70-100 points

Quizzes: 25-50 points

Project/Assignment: will vary

* Professional Development: 100 points

All percentages in this report will be rounded up to the nearest whole number.

* Professional Development allows you the opportunity to learn more about the things that are important to you and how they relate to the Marketing world. You are required to accumulate 100-150 points per nine weeks.

*If a student wishes to have an assignment re-graded (for any reason), they must fill out the Take 2 Form on the Google Classroom. I will only re-grade assignments when it is convenient for me to do so.

Sports And Entertainment Marketing Syllabus Barren

**Nevada. Department of Education.
Office of Career, Technical & Adult
Education**

Sports And Entertainment Marketing Syllabus Barren:

Embracing the Melody of Expression: An Mental Symphony within **Sports And Entertainment Marketing Syllabus Barren**

In a global consumed by monitors and the ceaseless chatter of instantaneous interaction, the melodic elegance and emotional symphony created by the published word often diminish into the background, eclipsed by the constant sound and distractions that permeate our lives. Nevertheless, situated within the pages of **Sports And Entertainment Marketing Syllabus Barren** a charming literary prize filled with organic feelings, lies an immersive symphony waiting to be embraced. Constructed by an elegant composer of language, that fascinating masterpiece conducts viewers on a psychological trip, well unraveling the hidden melodies and profound affect resonating within each carefully crafted phrase. Within the depths of the emotional analysis, we will explore the book is main harmonies, analyze its enthralling writing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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web choose 1 answer the final temperature is closer to t₁ than to t₂ a the final temperature is closer to t₁ than to t₂ the final temperature is exactly halfway between t₁ and t₂ b the final temperature is exactly halfway between t₁ and t₂ the final temperature is closer to t₂ than to t₁ c

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web in this set of practice questions we will go over the main types of questions on calorimetry including the heat capacity the heat of reaction finding the final temperature of a mixture constant pressure calorimetry and constant volume calorimetry a few important concepts and formulas you will need to solve these questions

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water calculate the molar heat of combustion of hexane if the water temperature rises 55.4 °C the specific heat capacity of water is 4.184 J/g °C h ms t h 5.650 grams H₂O 4.184 J/g °C h 1310 kJ

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web 6 a 218 g sample of steam at 121°C is cooled to ice at 14°C find the change in heat content of the system 7 in going from ice at 34°C to steam at 138°C a sample of water absorbs 1.41×10^5 J find the mass of the sample answers 1.268×10^4 J 2.623×10^4 J 3.171×10^4 J 4.111×10^5 J 5.211×10^5 J 6

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