

Sea-Salt Lab Report

Introduction:

Humankind has been producing salt since ancient times; as early as several hundred B.C., ancient civilizations had set up massive mines to pull the mineral from the ground. It was not long after that the distillation of salt from sea water began, a practice which has been continued to this day. The ancient's infatuation with salt has been explored in depth; it served as a preservative, medical aide (Aztec warriors would bathe themselves in brine, believing (correctly) that this caused their wounds to heal more quickly and cleanly) and condiment. Salt has weathered the ages as a valuable resource, and is widely consumed to this day (a 2010 study indicated that the average global consumption for salt was 10 grams per person per day).

Although there is much concern over people consuming too much sodium, what must not be forgotten is that salt (specifically the sodium cation) plays a vital role in the human body. The sodium cation serves as an electrolyte in the body, carrying electric charge and maintaining the balance of water between cells. A deficiency in sodium (called hyponatremia) can lead to symptoms ranging from a feeling of weakness to severe vomiting.

The following lab report concerns an experiment done to determine the percent composition of ocean water. The lab extracts excess materials from the water sample, then removes the salt from the sample by evaporating the water. By the end of this lab, we will have determined approximately how much salt is in a given sample of sea water.

Materials:

1. 1 Glass Erlenmeyer flask
2. 1 Glass beaker
3. 1 Sample sea water

Unknown Salt Lab Report

VM Jensen

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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