

PRACTICE CHEMISTRY PLACEMENT EXAM

1. How many of the numbers below have 5 significant figures?

0.0054

19.000

0.00006

1.6090×10^5

13607

- a) 1
b) 2
c) 3
d) 4
e) 5
2. Nitric acid is a solution of which of the following dissolved in water?
- a) HNO_4
b) H_2NO_3
c) HNO_2
d) HNO_3
e) H_2NO
3. Which of the following is the smallest mass?
- a) 2.1 kg
b) 4.2×10^{10} ng
c) 5.8×10^2 g
d) 8.4×10^4 cg
e) 6.7×10^4 mg
4. If 250 mL of a 0.50 M NaCl solution is diluted to 840 mL, what is the molarity of the resulting solution?
- a) 0.15 M
b) 6.7 M
c) 0.60 M
d) 1.7 M
e) 0.0025 M
5. Iron(III) sulfite has the formula
- a) Fe_2SO_3
b) $\text{Fe}_2(\text{SO}_4)_3$
c) $\text{Fe}_2(\text{SO}_3)_3$
d) Fe_2SO_4
e) Fe_2SO_3

Uic Chemistry Placement Test Practice

Roman Wölfel



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Table of Contents Uic Chemistry Placement Test Practice

1. Understanding the eBook Uic Chemistry Placement Test Practice
 - The Rise of Digital Reading Uic Chemistry Placement Test Practice
 - Advantages of eBooks Over Traditional Books
2. Identifying Uic Chemistry Placement Test Practice
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Uic Chemistry Placement Test Practice
 - User-Friendly Interface
4. Exploring eBook Recommendations from Uic Chemistry Placement Test Practice
 - Personalized Recommendations
 - Uic Chemistry Placement Test Practice User Reviews and Ratings
 - Uic Chemistry Placement Test Practice and Bestseller Lists
5. Accessing Uic Chemistry Placement Test Practice Free and Paid eBooks
 - Uic Chemistry Placement Test Practice Public Domain eBooks
 - Uic Chemistry Placement Test Practice eBook Subscription Services
 - Uic Chemistry Placement Test Practice Budget-Friendly Options
6. Navigating Uic Chemistry Placement Test Practice eBook Formats
 - ePub, PDF, MOBI, and More
 - Uic Chemistry Placement Test Practice Compatibility with Devices
 - Uic Chemistry Placement Test Practice Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Uic Chemistry Placement Test Practice
 - Highlighting and Note-Taking Uic Chemistry Placement Test Practice
 - Interactive Elements Uic Chemistry Placement Test Practice

8. Staying Engaged with Uic Chemistry Placement Test Practice
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Uic Chemistry Placement Test Practice
9. Balancing eBooks and Physical Books Uic Chemistry Placement Test Practice
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Uic Chemistry Placement Test Practice
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Uic Chemistry Placement Test Practice
 - Setting Reading Goals Uic Chemistry Placement Test Practice
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Uic Chemistry Placement Test Practice
 - Fact-Checking eBook Content of Uic Chemistry Placement Test Practice
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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