# Final Exam Review CHM 2045 questions with answers

element - \*\*□ VERIFIED ANSWERS□ ✓✓ simplest form of matter; cannot be broken down by chemical means. An atom is the smallest unit of an element.

A substance that cannot be broken down into simpler substances

compound - \*\*□ VERIFIED ANSWERS□ </made up of two or more elements chemically combined in fixed proportions. A molecule is the smallest unit of a compound.

mixture - \*\* UERIFIED ANSWERS - < made of substances (elements and/or compounds) mixed together in no fixed composition. A homogeneous mixture will appear uniform throughout is sometimes called a solution. A heterogeneous mixture varies in its composition (you will see distinct parts or phases).

Density - \*\*□ VERIFIED ANSWERS□ - For irregularly-shaped solids:

- 1- take mass of solid (or given)
- 2- determine the volume of solid by displacement of water

Example: A solid weighing 25.22 grams is dropped into a graduated cylinder containing 15.6 mL of water. The water level rises to 25.0 mL. Calculate the density of the solid.

D = Mass = 25.22 g = 25.22 g = 2.7 g/mLVolume (25.0 - 15.6)mL 9.4 mL

# **Uf Chm 2045 Final Exam**

# **Michael Brown**

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