

- 1 The table below gives the electron distributions of atoms of different elements.

element	electron distribution
A	2 + 7
B	2 + 8 + 4
C	2 + 8 + 8 + 1
D	2 + 8 + 18 + 5
E	2 + 8 + 18 + 7
F	2 + 8 + 18 + 18 + 6

For each of the following, select an element or elements from the table that matches the description. Each element may be selected once, more than once or not at all.

- (a) These two elements are in the same group.

_____ [1]

- (b) This element forms a fluoride with a formula of the type XF_2 .

_____ [1]

- (c) This element reacts violently with cold water.

_____ [1]

- (d) This element has a macromolecular structure similar to that of diamond.

_____ [1]

- (e) The only oxidation state of this element is 0.

_____ [1]

- (f) This element is bromine.

_____ [1]

- (g) This element is a good conductor of electricity.

_____ [1]

[Total: 7]

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