



11 How does a substance change from solid to liquid or gas? What is the change of state called? Can you give examples of substances that can do this?

10 Complete the diagram to show the arrangement of the particles in each state of matter. Does the particles under the action of the change of state.

9 What is a change of state, and what does it mean?

8 What property of matter is demonstrated by the different states of matter? Give examples of the different states of matter.

7 How do you know that the particles are moving?

PARTICLE THEORY AND STATES OF MATTER

What is the arrangement of particles in each state of matter? What is the change of state called? Can you give examples of substances that can do this?

Particle Theory is the theory that all matter is made up of tiny particles. These particles have different properties of particles in matter.

States of Matter is the name of the different states of matter. There are three states of matter: solid, liquid, and gas. The particles in each state are arranged differently.

Change of State is the name of the change from one state of matter to another. There are three types of change of state: melting, freezing, and boiling.

Solid State Particles are closely packed together. They have a fixed shape and volume. The particles in a solid are arranged in a regular pattern.

Liquid State Particles are closely packed but can move past each other. They have a fixed volume but no fixed shape. The particles in a liquid are arranged in a disordered pattern.

Gaseous State Particles are far apart and move freely. They have no fixed shape or volume. The particles in a gas are arranged in a disordered pattern.

Change of State Particles in a solid can change to a liquid or a gas. This is called melting or boiling. Particles in a liquid can change to a solid or a gas. This is called freezing or condensation. Particles in a gas can change to a liquid or a solid. This is called condensation or deposition.

Particle Theory explains the change of state. When a solid melts, the particles gain energy and move past each other. When a liquid boils, the particles gain energy and move away from the liquid. When a gas condenses, the particles lose energy and move closer together.

Change of State Particles in a solid can change to a liquid or a gas. This is called melting or boiling. Particles in a liquid can change to a solid or a gas. This is called freezing or condensation. Particles in a gas can change to a liquid or a solid. This is called condensation or deposition.

Particle Theory explains the change of state. When a solid melts, the particles gain energy and move past each other. When a liquid boils, the particles gain energy and move away from the liquid. When a gas condenses, the particles lose energy and move closer together.



6 How do you know that the particles are moving?

States Of Matter Guided Reading Answer Key

J Elliott



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Handbook of CALL Teacher Education and Professional Development Dara Tafazoli, Michelle Picard, 2023-04-26 This comprehensive handbook provides an overview of current trends in computer assisted language learning CALL teacher education and professional development across the globe It highlights theories and practices in CALL teacher education and professional development in five sections such as English language teaching including pre service teachers in service teachers teacher educators material developers course designers and

researchers It explores the role of CALL teacher education and professional development in many underexplored countries such as Africa Asia Eastern Europe and the Middle East It stresses the critical role of professional development programs from the use of technology in its generic sense The theoretical and empirical chapters in the book provide a more inclusive and comprehensive picture of various aspects of CALL teacher education and professional development globally It offers context specific approaches and strategies to language teachers and teacher educators It provides pedagogical implications and suggestions for promoting digital literacy and autonomy in online education This book provides valuable insights for researchers teacher educators and teacher trainers in applied linguistics Prentice Hall Chemistry ,2000 *Properties of Matter Gr. 5-8* George Graybill,2007-09-01 Discover what matter is and what it isn't Our resource breaks down the physical and chemical properties of matter to make it more accessible to students Start off by identifying matter as atoms particles and molecules Then explore the three states of matter solid liquid and gas Determine whether something is transparent opaque or translucent List three physical changes and three chemical changes that could happen in the kitchen Conduct an experiment to see chemical change in action Describe the steps necessary when separating a mixture Experiment with photosynthesis an important chemical change Aligned to the Next Generation Science Standards and written to Bloom's Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included

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